

14

THE BEHAVIOR OF GASES

Chapter Test II

A. Matching

Match a word or phrase in Column I with the correct description in Column II. Write the letter of the correct answer next to the item.

Column I

1. The constant relating unit temperature change to the change in volume of a fixed mass of gas at constant pressure
2. The volume of a fixed mass of gas at constant pressure as the absolute temperature is changed
3. The ratio of the mass of a gas to its volume, independent of the state of the gas
4. The constant that is the product of the pressure and volume of a gas
5. A constant of nature that has the value of about 8.314 J/mol·K
6. The a parameter of gas molecules, measured for monatomic gas molecules only
7. The constant that is the product of the pressure and volume of a gas
8. $\frac{P_1 V_1}{T_1} = \frac{P_2 V_2}{T_2}$
9. The amount of heat enough to raise 1 g of water 1°C
10. A unit of energy

Column II

- a. pressure-volume work
- b. ideal gas constant R
- c. density
- d. compressibility
- e. Boyle's law
- f. molar volume
- g. constant-volume molar heat capacity
- h. volume
- i. compressibility
- j. heat capacity of a gas

B. Multiple Choice

Choose the correct answer and write its letter on the line.

1. According to kinetic theory, the average kinetic energy of a gas molecule is proportional to the temperature of the gas.
a. the absolute temperature
b. the Celsius temperature
c. the Fahrenheit temperature
d. the Kelvin temperature

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