

9-2 Practice Problems (Continued)

11. Write a balanced chemical equation for the reaction in which Dinitrogen Pentoxide reacts with water to produce Nitric Acid.
12. Magnesium reacts with Titanium (IV) Chloride to produce Magnesium Chloride and Titanium. Write the balanced equation for this reaction.
13. Write a balanced chemical equation for the reaction in which Carbon reacts with Zinc Oxide to produce Zinc and Carbon Dioxide.
14. Bromine reacts with Sodium Iodide to form Sodium Bromide and Iodine. Write the balanced equation for this reaction.
15. Write a balanced chemical equation for the reaction in which phosphorus Trichloride reacts with Chlorine gas to produce Phosphorus Pentachloride.
16. Phosphorus reacts with Bromine to produce Phosphorus Tribromide. Write the balanced equation for this reaction.
17. Calcium Hydride reacts with water to produce Calcium Hydroxide and Hydrogen gas. Write the balanced equation for this reaction.
18. Write a balanced chemical equation for the reaction in which Sulfuric Acid reacts with Potassium Hydroxide to produce Potassium Sulfate and water.
19. Write a balanced chemical equation for the reaction in which Propane (C_3H_8) reacts with Oxygen gas to produce Carbon Dioxide and water.
20. Benzene (C_6H_6) reacts with Oxygen gas to produce Carbon Dioxide and water. Write the balanced equation for this reaction.

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Davis Baird



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Problems and Problem Solving in Chemistry Education Georgios Tsaparis, 2021-05-17 Problem solving is central to the teaching and learning of chemistry at secondary tertiary and post tertiary levels of education opening to students and professional chemists alike a whole new world for analysing data looking for patterns and making deductions As an important higher order thinking skill problem solving also constitutes a major research field in science education Relevant education research is an ongoing process with recent developments occurring not only in the area of quantitative computational problems but also in qualitative problem solving The following situations are considered some general others with a focus on specific areas of chemistry quantitative problems qualitative reasoning metacognition and resource activation deconstructing the problem solving process an overview of the working memory hypothesis reasoning with the electron pushing formalism scaffolding organic synthesis skills spectroscopy for structural characterization in organic chemistry enzyme kinetics problem solving in the academic chemistry laboratory chemistry problem solving in context team based active learning technology for molecular representations IR spectra simulation and computational quantum chemistry tools The book concludes with methodological and epistemological issues in problem solving research and other perspectives in problem solving in chemistry With a foreword by George Bodner **Addison-Wesley Chemistry** Antony C. Wilbraham, Dennis D. Staley, Michael S. Matta, Edward L. Waterman, Prentice-Hall Staff, 2001-02 To purchase or download a workbook click on the Purchase or Download button to the left To purchase a workbook enter the desired quantity and click Add to Cart To download a free workbook right click the FREE Download PDF link and save to your computer This will result in a faster download as opposed to left clicking and opening the link *Chemistry Education* Javier García-Martínez, Elena Serrano-Torregrosa, 2015-05-04 Winner of the CHOICE Outstanding Academic Title 2017 Award This comprehensive collection of top level contributions provides a thorough review of the vibrant field of chemistry education Highly experienced chemistry professors and education experts cover the latest developments in chemistry learning and teaching as well as the pivotal role of chemistry for shaping a more sustainable future Adopting a practice oriented approach the current challenges and opportunities posed by chemistry education are critically discussed highlighting the pitfalls that can occur in teaching chemistry and how to circumvent them The main topics discussed include best practices project based education blended learning and the role of technology including e learning and science visualization Hands on recommendations on how to optimally implement innovative strategies of teaching chemistry at university and high school levels make this book an essential resource for anybody interested in either teaching or learning chemistry more effectively from experience chemistry professors to secondary school teachers from educators with no formal training in didactics to frustrated chemistry

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