

Section Review 8.1

Part A Completion

1. compound
2. molecular
3. nonmetals
4. diatomic
5. molecular formula
6. low
7. high
8. atoms
9. structure

Part B True-False

10. ST
11. NT
12. AT
13. NT
14. ST

Part C Matching

15. c
16. a
17. d
18. b
19. c

Part D Questions and Problems

20. molecular compound
21. a. 4 carbon atoms, 10 hydrogen atoms
b. 6 carbon atoms, 5 hydrogen atoms, 1 fluorine atom
22. a. molecule
b. atom
c. molecule
d. atom
e. molecule

Section Review 8.2

Part A Completion

1. stable electron
2. covalent
3. shared
4. single
5. unshared pairs
6. double/triple
7. coordinate covalent bond
8. Energy
9. bond dissociation energy
10. resonance structure

Part B True-False

11. NT
12. NT
13. AT
14. AT
15. ST
16. NT

Part C Matching

17. c
18. a
19. b
20. c
21. d

Part D Questions and Problems

22. a. $\text{Br}:\text{Br}:$
- b. $\text{H}:\text{C}::\text{N}:$
- c. $\begin{array}{c} \text{H} \\ \diagup \quad \diagdown \\ \text{H}:\text{N}:\text{H} \\ | \\ \text{H} \end{array}$

Section 8.3

Part A Completion

1. molecular orbitals
2. bonding orbital
3. lower
4. sigma or σ
5. pi or π
6. three-dimensional
7. VSEPR theory
8. orbital hybridization

Part B True-False

9. AT
10. ST
11. NT
12. AT
13. ST
14. AT

Part C Matching

15. d
16. c
17. a
18. b
19. c

Part D Questions and Problems

20. Reading from left to right:
 sp^3 sp^2 sp^2 sp sp sp^3

Section 8.4

Part A Completion

1. equally
2. nonpolar
3. unequally
4. polar
5. electronegativities
6. dipole interactions
7. hydrogen bond
8. electronegative
9. oxygen, nitrogen, or fluorine

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K Morrison

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