

Name _____ Date _____ Class _____

17

THERMOCHEMISTRY

SECTION 17.1 THE FLOW OF ENERGY—HEAT AND WORK (pages 505–510)

This section explains the relationship between energy and heat, and distinguishes between heat capacity and specific heat.

► Energy Transformations (page 505)

- What area of study in chemistry is concerned with the heat transfers that occur during chemical reactions? thermochemistry
- Where the use of energy is concerned (in a scientific sense), when is work done?
Work is done when a force is used to move an object.
- Circle the letter next to each sentence that is true about energy:
 - ☒ Energy is the capacity for doing work or supplying heat.
 - ☒ Energy is detected only because of its effects.
 - ☐ Heat is energy that transfers from one object to another because they are at the same temperature.
 - ☒ Gasoline contains a significant amount of chemical potential energy.
- Circle the letter next to each sentence that is true about heat.
 - ☒ One effect of adding heat to a substance is an increase in the temperature of that substance.
 - ☐ Heat always flows from a cooler object to a warmer object.
 - ☒ If two objects remain in contact, heat will flow from the warmer object to the cooler object until the temperature of both objects is the same.

► Exothermic and Endothermic Processes (pages 506–507)

- What can be considered the “system” and what are the “surroundings” when studying a mixture of chemicals undergoing a reaction? Write your answers where indicated below.

System: The mixture of chemicals itself is considered the system.

Surroundings: Everything but the mixture of chemicals is the surroundings, but practically speaking, the immediate vicinity of the system.

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Lei Huang



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to left clicking and opening the link **Computational Chemistry** Errol G. Lewars,2016-09-20 This is the third edition of the successful text reference book that covers computational chemistry It features changes to the presentation of key concepts and includes revised and new material with several expanded exercises at various levels such as harder questions for those ready to be tested in greater depth this aspect is absent from other textbooks in the field Although introductory and assuming no prior knowledge of computational chemistry it covers the essential aspects of the subject There are several introductory textbooks on computational chemistry this one is as in its previous editions a unique textbook in the field with copious exercises and questions and solutions with discussions Noteworthy is the fact that it is the only book at the introductory level that shows in detail yet clearly how matrices are used in one important aspect of computational chemistry It also serves as an essential guide for researchers and as a reference book *Physical Chemistry and Its Biological Applications* Wallace Brey,2012-12-02 Physical Chemistry and Its Biological Applications presents the basic principles of physical chemistry and shows how the methods of physical chemistry are being applied to increase understanding of living systems Chapters 1 and 2 of the book discuss states of matter and solutions of nonelectrolytes Chapters 3 to 5 examine laws in thermodynamics and solutions of electrolytes Chapters 6 to 8 look at acid base equilibria and the link between electromagnetic radiation and the structure of atoms Chapters 9 to 11 cover different types of bonding the rates of chemical reactions and the process of adsorption Chapters 12 to 14 present molecular aggregates magnetic resonance spectroscopy and photochemistry and radiation This book is useful to biological scientists for self study and reference With modest additions of mathematical material by the teacher the book should also be suitable for a full year major s course in physical chemistry A TEXTBOOK OF CHEMICAL ENGINEERING THERMODYNAMICS K. V. NARAYANAN,2013-01-11 Designed as an undergraduate level textbook in Chemical Engineering this student friendly thoroughly class room tested book now in its second edition continues to provide an in depth analysis of chemical engineering thermodynamics The book has been so organized that it gives comprehensive coverage of basic concepts and applications of the laws of thermodynamics in the initial chapters while the later chapters focus at length on important areas of study falling under the realm of chemical thermodynamics The reader is thus introduced to a thorough analysis of the fundamental laws of thermodynamics as well as their applications to practical situations This is followed by a detailed discussion on relationships among thermodynamic properties and an exhaustive treatment on the thermodynamic properties of solutions The role of phase equilibrium thermodynamics in design analysis and operation of chemical separation methods is also deftly dealt with Finally the chemical reaction equilibria are skillfully explained Besides numerous illustrations the book contains over 200 worked examples over 400 exercise problems all with answers and several objective type questions which enable students to gain an in depth understanding of the concepts and theory discussed The book will also be a useful text for students pursuing courses in chemical engineering related branches such as polymer engineering petroleum engineering and safety and environmental

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