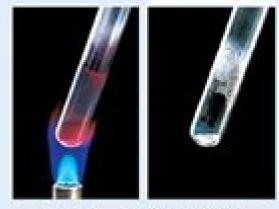
## 2 Decomposition Reaction

Reactants: Generally a single binary compound or a compound with a polyatomic ion Probable Products: Two elements (for a binary compound), or two or more elements and/or compounds (for a compound with a polyatomic ion)

Example: Heating mercury(II) oxide



$$2HgO(s) \longrightarrow 2Hg(I) + O_2(g)$$



# **Prentice Hall Chemistry Answer Chemical Reactions**

**Clifford Lane** 

**Prentice Hall Chemistry Answer Chemical Reactions:** 

This book delves into Prentice Hall Chemistry Answer Chemical Reactions. Prentice Hall Chemistry Answer Chemical Reactions is a crucial topic that needs to be grasped by everyone, from students and scholars to the general public. The book will furnish comprehensive and in-depth insights into Prentice Hall Chemistry Answer Chemical Reactions, encompassing both the fundamentals and more intricate discussions.

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