# Predicting Reaction Products

Predict the products for the following reactions, balance the equation, then classify the type of reactions:

- \_\_\_Na+\_\_FeBr₃→ 1)
- \_\_\_ NaOH + \_\_\_ H₂SO4 → 2)
- 3) C<sub>2</sub>H<sub>4</sub>O<sub>2</sub> + O<sub>2</sub> →
- NH<sub>3</sub> + H<sub>2</sub>O → 40
- 5) PbSO. + AgNO<sub>3</sub> →
- 60  $PBr_s \rightarrow$
- 7) HBr+ Fe→
- 8)  $_{\text{L}}$  KMnO<sub>4</sub> +  $_{\text{L}}$  ZnCl<sub>2</sub>  $\rightarrow$
- 99  $MnO_2 + Sn(OH)_4 \rightarrow$
- O₂ + C₂H₂₂O₂ → 10)
- 11) H₂O₂ →
- 12) PtCl<sub>4</sub> + Cl<sub>2</sub> →

# **Predicting Products Packet Key**

**Arthur Winter** 

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