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PRECIPITATION REACTION LAB

Introduction:

The majority of ionic solids like salts (KCl and NaCl) are soluble in water because the polar water molecules surround the individual ions of the salt. Those that do not dissolve and go into solution form solid products called precipitates. These precipitates have many colors and often help scientists identify what the precipitate is present. In meteorology the term precipitation refers to meteorological phenomena such as rain or snow. Precipitation is chemical solutions occurs when two chemicals react to form a product that is insoluble in water and falls out of solution like rain or snow. A precipitate is a solid substance that separates from solution during a chemical reaction. A precipitate can be identified by the cloudy, milky, gelatinous, or grainy appearance it gives to the mixture.

A barium sulfate precipitate can be produced by the reaction of barium chloride and sodium sulfate. A chemical equation to describe the reaction is written and balanced like this:

$$BaCl_{2 (aq)} + Na_2SO_{4 (aq)} \rightarrow 2 NaCl_{(aq)} + BaSO_4(s)$$

Barium sulfate, BaSO₄ (s), is a common precipitate used as an X-ray contrast medium because it is insoluble in water an opaque to x-rays. Typically a patient drinks an aqueous slurry of barium sulfate just before he is x-rayed. The precipitate coats his stomach and intestines. These organs then show up on the x-ray film in vivid contrast, aiding the doctor's diagnosis. Notice the reaction that forms BaSO₄ is a double-replacement reaction in which the cations and anions of

the reactants trade partners to form the products.

To write a total ionic equation, rewrite all aqueous substances as their component ions and keep all solid substances unchanged.

$$Ba^{+2}_{(aq)} + 2Cl^{-}_{(aq)} + 2Na^{+}_{(aq)} + SO_4^{-2}_{(aq)} \rightarrow 2Na^{+}_{(aq)} + 2Cl^{-}_{(aq)} + BaSO_{4(s)}$$

Cancel all components of the reaction that are identical on both the reactant and product side of the reaction. These are called spectator ions. Write the net ionic equation by rewriting the reaction equation including only those reactants and products that change in the reaction.

Precipitation Reactions Lab Report Answers

Daniel F McAuley

Precipitation Reactions Lab Report Answers:

Unveiling the Power of Verbal Art: An Psychological Sojourn through **Precipitation Reactions Lab Report Answers**

In some sort of inundated with displays and the cacophony of immediate communication, the profound energy and psychological resonance of verbal artistry often diminish in to obscurity, eclipsed by the regular barrage of sound and distractions. Yet, nestled within the musical pages of **Precipitation Reactions Lab Report Answers**, a fascinating function of literary beauty that impulses with natural feelings, lies an memorable journey waiting to be embarked upon. Published by a virtuoso wordsmith, that exciting opus instructions viewers on an emotional odyssey, delicately revealing the latent possible and profound affect embedded within the complicated web of language. Within the heart-wrenching expanse of this evocative analysis, we shall embark upon an introspective exploration of the book is central themes, dissect its interesting publishing fashion, and immerse ourselves in the indelible impact it leaves upon the depths of readers souls.

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