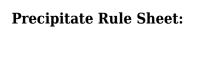
Table 17.3 Solubilities of lonic Compounds* aq = aqueous (dissolves in water); s = solid (does not dissolve in water)

lons	Acetate	Bromide	Carbonate	Chlorate	Chloride	Fluoride	Hydrogen Carbonate	Hydroxide	lodide	Nitrate	Nitrite	Phosphate	Sulfate	Sulfide	Sulfite
Aluminum	S	aq		aq	aq	s		S	_	aq		s	aq	-	
Ammonium	aq	aq	aq	aq	aq	aq	aq	_	aq	aq	aq	aq	aq	aq	aq
Barium	aq	aq	s	aq	aq	S		aq	aq	aq	aq	s	s		s
Calcium	aq	aq	S	aq	aq	s		s	aq	aq	aq	s	s	-	s
Cobalt(II)	aq	aq	S	aq	aq	-	i i	S	aq	aq	i n	s	aq	S	s
Copper(II)	aq	aq	S	aq	aq	aq		S		aq		S	aq	S	
Iron(II)	aq	aq	s		aq	s		S	aq	aq		s	aq	s	s
Iron(III)	-	aq			aq	s		s	aq	aq		s	aq	_	
Lead(II)	aq	S	S	aq	s	s	i i	S	s	aq	aq	s	s	s	s
Lithium	aq	aq	aq	aq	aq	aq	aq	aq	aq	aq	aq	S	aq	aq	aq
Magnesium	aq	aq	s	aq	aq	s		s	aq	aq	aq	S	aq	-	aq
Nickel	aq	aq	s	aq	aq	aq	J j	s	aq	aq		s	aq	S	s
Potassium	aq	aq	aq	aq	aq	aq	aq	aq	aq	aq	aq	aq	aq	aq	aq
Silver	s	S	S	aq	S	aq			s	aq	s	s	s	S	s
Sodium	aq	aq	aq	aq	aq	aq	aq	aq	aq	aq	aq	aq	aq	aq	aq
Zinc	aq	aq	8	aq	aq	aq		S	aq	aq		s	aq	s	s

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