

Write Titrating Volume (Spreads: 1.00)

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Acetic Acid (pKa = 4.75) titrated with NaOH

a. 0.100 M / volume 10.00 mL / acid / base 0.100 M

b. 0.100 M / volume 10.00 mL / acid / base 0.100 M

c. 0.100 M / pH 4.75 / compound that releases acetate
0.100 M / acid / base 0.100 M / acid

Acetic Acid: 1st, 2nd, 3rd, 4th, 5th, 6th, 7th, 8th, 9th, 10th

0.100 M / 0.100 M / 0.100 M / 0.100 M / 0.100 M / 0.100 M / 0.100 M / 0.100 M / 0.100 M / 0.100 M

Species	Acid (pH, Base (pH, or both (pH)	Acids with (pH, or both (pH)	Conductivity (C or NaC)
CH ₃ COOH	4.75	4.75	4.75
CH ₃ COO ⁻	4.75	4.75	4.75
NaOH	14.00	14.00	14.00
NaCl	7.00	7.00	7.00

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Raffaela Di Napoli



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