

Types of Chemical Reactions – Complete the note taking process for the example as we study chemical reactions.

1. **Synthesis Reactions** – two or more substances combine to form a new compound. (eg. $4\text{Fe} + 3\text{O}_2 \rightarrow 2\text{Fe}_2\text{O}_3$)

Symbol Equation: $\text{A} + \text{B} \rightarrow \text{AB}$

Example Equation: $2\text{H}_2(\text{g}) + \text{O}_2(\text{g}) \rightarrow 2\text{H}_2\text{O}(\text{g})$

2. _____ Reaction –

Symbol Equation:

Example Equation:

3. _____ Reaction –

Symbol Equation:

Example Equation:

4. _____ Reaction –

Symbol Equation:

Example Equation:

5. _____ Reaction –

Symbol Equation:

Example Equation:

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