

Instructions:-

Question paper consist of 31 questions divided into four sections namely A, B, C, D.

- 1) **Section-A:** Q. No. 1 contains ten multiple choice types of questions carrying one mark each.
Q.No.2 contains eight very short answer types of questions carrying one mark each.
- 2) **Section-B :** Q.No.3 to Q. No. 14 are Twelve questions carrying two mark each out of which any eight to be attempted.
- 3) **Section-C :** Q. No.15 to Q.No.26 are Twelve questions carrying three mark each out of which any eight are to be attempted.
- 4) **Section-D :** Q.No.27 to Q.No. 31 are Five questions carrying four mark each out of which any three to be attempted.

Section- A

Q.1) Select and write the correct answer:

(10)

- i. Which of the following compounds can NOT demonstrate the law of proportions?
a) NO, NO₂ b) CO, CO₂ c) H₂O, H₂O₂ d) Na₂S, NaF
- ii. Which of the following temperature will read the same value on Celsius and Fahrenheit Scales
a) -40 b) + 40 c) -80 d) -20
- iii. SI unit of the quantity electric current is
a) Volts b) Ampere c) Candela d) Newton
- iv. The no. of molecules in 22.4 cm³ of Nitrogen Gas at STP is
a) 6.022×10^{20} b) 6.022×10^{23} c) 22.4×10^{20} d) 22.4×10^{23}
- v. The homogeneous series of alcohol has general molecular formula ?
a) C_n H₂n+2OH b) C_n H₂n OH c) C_n H₂n+1OH d) none
- vi. Find the molar mass of CH₃CHO.
a) 44 b) 40 c) 80 d) 30
- vii. Find the molar mass of CH₃NO₂.
a) 61 b) 77 c) 98 d) 51
- viii. Atomic number of an elements is equal to
a) P = e- b) p = N c) P = A d) p = n = e

Nmaths 2examination Paper

J Elliott



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