

Polycom® VoiceStation™ 100

Affordable quality sound for smaller rooms or desktops



Highlights

Acoustic Clarity Technology – Allows simultaneous natural conversation – the industry's best full duplex.

360-degree room coverage – A powerful, digitally-tuned custom speaker and three sensitive microphones provide uniform coverage of a small conference room or office.

Easy-to-install and use – Start conferencing in minutes!

The Polycom VoiceStation 100 provides a natural hands-free business communication solution. With Polycom's Acoustic Clarity Technology, the VoiceStation 100 lets you have productive, two-way conversations.

Handset speakerphones are fine for listening to voicemail or conducting short conversations, but they usually don't offer the technology that makes natural two-way communication possible.

Polycom's voice conferencing systems are found in more conference rooms, boardrooms, and other meeting areas than all other types of voice conferencing systems combined. The affordable VoiceStation 100 provides basic conferencing for small rooms and offices.

For business-critical clarity, there's one clear choice: VoiceStation 100.

The affordable, basic conferencing choice for The Polycom Office™

With integrated video, voice, data, and Web capabilities, The Polycom Office is the only solution that offers you an easy way to connect, conference, and collaborate any way you want. The Polycom Office is our commitment to making distance communications as natural and interactive as being there. Work faster, smarter, and better with the VoiceStation 100 and The Polycom Office.



Voice

Connect. Any Way You Want.

Polycom Voicestation 100 Manual

R Barnett



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web with a strong acid and strong base titration strong acids and strong bases refer to species that completely dissociate to form ions in solution by contrast weak acids and bases ionize only partially and the ionization reaction is reversible

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web by utilizing the half equivalence point on the titration curve the pka of the acid analyte was determined to be 3.655 furthermore using the stoichiometric ratio between the naoh and the acid analyte 1:1 we could experimentally determine the number of moles of the acid in the solution to be 0.00731 moles making the experimental molar mass

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web titration experiment in the neutralization of hydrochloric acid by sodium hydroxide the mole ratio of acid to base is 1:1
 $\text{HCl(aq)} + \text{NaOH(aq)} \rightarrow \text{NaCl(aq)} + \text{H}_2\text{O(l)}$
 one mole of HCl would be fully neutralized by one mole of NaOH

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web specifically an acid base titration can be used to figure out the concentration of an acid or base whether an unknown acid or base is strong or weak pka of an unknown acid or pkb of the unknown base key terms titration a process where a solution of known strength is added to a certain volume of a treated sample containing an

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web this point of titration is referred to as the equivalence point considering the 1:1 stoichiometry of this acid base reaction NaOH aq $\text{C}_6\text{H}_4\text{COOH}$ aq $\text{C}_6\text{H}_4\text{COONa}$ aq H_2O l the point of equivalence is the point of titration when the number of moles of naoh added is equal to the number of moles of khp in the