## Naming Compounds Containing Polyatomic Ions

Use the following list of polyatomic ions to name the compounds given.

	OH NO <sub>2</sub> C <sub>2</sub> H <sub>2</sub> O <sub>2</sub>	hydroxide nitrate acetate	CIO <sub>3</sub> HCO <sub>3</sub> SO <sub>4</sub> CO <sub>3</sub>	chlorate bicarbonate sulfate carbonate	SO, PO, NH,	sulfite phosphate ammonium
1.	CuSO			26. NaOH		
22.	KCIO,			27. NaNO,		
3.	NH4OH	+1				
4.	K2CO3					
5	5. Na <sub>3</sub> SO <sub>4</sub>			SO. LiOH		
6.	KC <sub>2</sub> H <sub>2</sub> O <sub>2</sub>			31. KHCO,		
7.	. NH,CI			32. NH,F		
8.	Mg(OH) <sub>2</sub>					
9.	$Zn(NO_3)_2$			6. NaC <sub>2</sub> H <sub>3</sub> O <sub>2</sub>		
10.	MgSO					
11.	CaCO <sub>a</sub>					
12.	кон					
13.	Ca(OH) <sub>2</sub>					
1.4.	K <sub>2</sub> CO <sub>3</sub>					
15.	AIPO					
16.	. Na <sub>3</sub> PO <sub>4</sub>			41. NH,I		
17.	. MgSO <sub>4</sub>			42. NH, HCO,		
18.	AgNO <sub>3</sub>			43. Cu <sub>3</sub> (PO <sub>4</sub> ) <sub>2</sub>		
19.	Na <sub>3</sub> CO <sub>5</sub>			44. Ni()H		
	NaHCO <sub>3</sub>			45. ZnSO,		
21.	Ca(NO <sub>5</sub> ) <sub>2</sub>					
22.	BaSO,			47. HgOH		
24.	LiNO <sub>a</sub>					

# **Polyatomic Ions Naming Answer**

**AW Rasmussen** 

#### **Polyatomic Ions Naming Answer:**

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