Question		ion	Answer		Guidance
6	*	1	A is the (de Broglie) wavelength associated with a particle h is (a constant known as) the Planck constant mv is the momentum/mass x speed of the particle	81 81	allow object/electron for particle accept velocity for speed accept m and v separately but the particle must appear with one of the quantities to score the mark
Ī			electrons passing through a (thin) sheet of graphite are diffracted producing rings/pattern (on a fluorescent screen)	B1 A1	any suitable situation NOT Au or Al, etc. what is observed + interpretation 1 mark. Do not award the A mark alone unless a plausible situation has been suggested. Young slits type experiments etc score G/2
		m 1	$eV = 10 \text{ mV}^2$ $2eVVm = V^2$ $50 V = \sqrt{(2 \times 1.6 \times 10^{10} \times 5.0 \times 10^{5/9}.1 \times 10^{10})}$ $50 V = 1.3 \times 10^6 \text{ (m s}^3)$	C1 C1 B1 A0	allow eV = 8.0 x 10 ⁻⁶ J for 1 mark allow √(2 x 8.0 x 10 ⁻⁶ /9.1 X 10 ⁻⁶) or evidence of correct calculation, e.g. v = 1.325 x 10 ⁶
		HI 2	$A = h/mv = 6.63 \times 10^{-14} / 1.3 \times 10^{8} \times 9.1 \times 10^{10}$ $A = 5.6 \times 10^{-16} \text{ (m)}$	C1 A1	allow 5.5 × 10 12 if 1.325 × 10 ⁸ is used
		111.3	A of visible light 5.6 x 10 ° m so power of 10 = 5 or -5	81 81	ecf aiii2; accept 4 to 7 x 10 ⁷ m NOT 10 ⁷ NOT 10 on the answer line
	b		The photoelectric effect Individual photons are absorbed by individual electrons in the metal surface, i.e. one to one interaction. Only photons with energies above the work function energy will cause photoelectron emission/idea of threshold frequency. Hence u-v photons or blue photons will cause photoemission but red photons will not. Number of electrons emitted depends on light intensity. A wave model does not explain instantaneous emission of electrons. A wave model does not explain a threshold frequency/wavelength for emission to occur.	81 81 81 81 81	QWC mark allow exp't description: uv light shone on a zinc plate connected to a gle max 3 from 6 marking points allow wave model does not explain no emission however bright the light if energy of photon below work function
			Total question 6	16	

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Jin-Ying Zhang

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