

- 99) Ionization energies allow chemists to determine electron structure. The first two ionization energies of rubidium and strontium are given in the table.

element	first ionization energy (kJ/mol)	second ionization energy (kJ/mol)
rubidium	419	2633
strontium	550	1064

- 100) Write an equation to represent the **second** ionization energy of strontium.

Include state symbols.

[2]

- 101) Why is the **first** ionization energy of strontium larger than the **first** ionization energy of rubidium?

In your answer you should use appropriate chemical terms **correctly**.

[2]

- 102) Why is the **second** ionization energy of rubidium larger than the **second** ionization energy of strontium?

[2]

(Total: 12)

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**Lei Shi**

A red circular graphic with a gradient, appearing as a semi-circle or a partial circle, located to the right of the name 'Lei Shi'.

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### **Table of Contents Ocr Chemistry F321 Jan 13 Paper**

1. Understanding the eBook Ocr Chemistry F321 Jan 13 Paper
  - The Rise of Digital Reading Ocr Chemistry F321 Jan 13 Paper
  - Advantages of eBooks Over Traditional Books
2. Identifying Ocr Chemistry F321 Jan 13 Paper
  - Exploring Different Genres
  - Considering Fiction vs. Non-Fiction
  - Determining Your Reading Goals
3. Choosing the Right eBook Platform
  - Popular eBook Platforms
  - Features to Look for in an Ocr Chemistry F321 Jan 13 Paper
  - User-Friendly Interface
4. Exploring eBook Recommendations from Ocr Chemistry F321 Jan 13 Paper
  - Personalized Recommendations
  - Ocr Chemistry F321 Jan 13 Paper User Reviews and Ratings
  - Ocr Chemistry F321 Jan 13 Paper and Bestseller Lists

5. Accessing Ocr Chemistry F321 Jan 13 Paper Free and Paid eBooks
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  - Ocr Chemistry F321 Jan 13 Paper eBook Subscription Services
  - Ocr Chemistry F321 Jan 13 Paper Budget-Friendly Options
6. Navigating Ocr Chemistry F321 Jan 13 Paper eBook Formats
  - ePub, PDF, MOBI, and More
  - Ocr Chemistry F321 Jan 13 Paper Compatibility with Devices
  - Ocr Chemistry F321 Jan 13 Paper Enhanced eBook Features
7. Enhancing Your Reading Experience
  - Adjustable Fonts and Text Sizes of Ocr Chemistry F321 Jan 13 Paper
  - Highlighting and Note-Taking Ocr Chemistry F321 Jan 13 Paper
  - Interactive Elements Ocr Chemistry F321 Jan 13 Paper
8. Staying Engaged with Ocr Chemistry F321 Jan 13 Paper
  - Joining Online Reading Communities
  - Participating in Virtual Book Clubs
  - Following Authors and Publishers Ocr Chemistry F321 Jan 13 Paper
9. Balancing eBooks and Physical Books Ocr Chemistry F321 Jan 13 Paper
  - Benefits of a Digital Library
  - Creating a Diverse Reading Collection Ocr Chemistry F321 Jan 13 Paper
10. Overcoming Reading Challenges
  - Dealing with Digital Eye Strain
  - Minimizing Distractions
  - Managing Screen Time
11. Cultivating a Reading Routine Ocr Chemistry F321 Jan 13 Paper
  - Setting Reading Goals Ocr Chemistry F321 Jan 13 Paper
  - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Ocr Chemistry F321 Jan 13 Paper
  - Fact-Checking eBook Content of Ocr Chemistry F321 Jan 13 Paper
  - Distinguishing Credible Sources
13. Promoting Lifelong Learning

- Utilizing eBooks for Skill Development
  - Exploring Educational eBooks
14. Embracing eBook Trends
- Integration of Multimedia Elements
  - Interactive and Gamified eBooks

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