

### Chemical Formula

Talking about proportions, molecules and relative of atoms indicated by a subscript.

Molecular  $C_2H_6O$  (methylating)

Formula represents the number of \_\_\_\_\_ of each  
in a single molecule.

Empirical  $C_2H_4O$  (methylating)

Formula represents the simplest ratio of  
positive and negative ions in a formula.

### Atomic Mass

The average mass of atoms relative to hydrogen  
atoms together.

### Isotopes

$^{12}_6C$  stable isotope carbon

$^{13}_6C$  stable isotope carbon

### Radioisotopes

$^{14}_6C$  radioisotope carbon

$^{15}_6C$  radioisotope carbon

$^{16}_6C$  radioisotope

# Note Taking Guide Episode 6chemical Formula Key

**Patricia A. Potter, Anne G.  
Perry, Patricia A. Stockert, Amy Hall**

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