

Chemical Formula

Talking about proportions, molecules and relative all atoms present (in a compound)

Molecular $\text{C}_{10}\text{H}_{18}\text{O}$ (molecular)

Formula represents the number of atoms of each in a single molecule

Empirical $\text{C}_5\text{H}_9\text{O}$ (molecular)

Formula represents the simplest ratio of positive and negative ions

Atomic Mass

The average mass of atoms present together

Isotopes

^{12}C stable isotope abundant

^{13}C stable isotope abundant

Radioisotopes

^{14}C radioisotope abundant

^{15}C radioisotope abundant

^{16}C radioisotope

Note Taking Guide Episode 601 Physics

DJ Losen



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