

Faradaic current:

- i) contribution of the potential \_\_\_\_\_ energy of the half-cell  
in a sample of solution
- ii) does not depend on the amount of \_\_\_\_\_ in that sample
- iii) depends on: \_\_\_\_\_ used to decompose solution

Non-:

- i) \_\_\_\_\_ amount of \_\_\_\_\_ energy.  $n$  mol (from decrease of  
concentration in \_\_\_\_\_)
- ii) depends on \_\_\_\_\_ of solution.
- iii) depends on: \_\_\_\_\_ and on \_\_\_\_\_ ( $n$  &  $F = 96485 \text{ C/mol}$ )

Electro-energies:  
 Potential energy of \_\_\_\_\_  
 • Potential energy is being used to \_\_\_\_\_  
 \_\_\_\_\_  
 • Only \_\_\_\_\_ and it can be measured

Specific heat capacity:

- i) amount of \_\_\_\_\_ required to raise the \_\_\_\_\_ of  
 $1 \text{ g}$  of substance  $1 \text{ K/Celcius}$
- ii) depends on: \_\_\_\_\_ and on \_\_\_\_\_



We measure  $\Delta G$  to know how big energy will be for  $\Delta G$  gives some information on  
 whether \_\_\_\_\_ energy and part of it is used to make the molecules more ordered  
 (this) making the \_\_\_\_\_ [1+1]

they also need gas bubble so they don't collect at night that the water's  
 some structure being made from water

# Note Taking Guide Episode 1301 Chemistry Teacher Sheet

**Yi-Tong Ma**



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