

Name: Marcie Hall Period: 3 Date: 5/1/23



Note-Taking Guide (CER)

Activity Title: Aquatic Ecosystems- Developing

Activity Scoreboard when finished, circle highest score achieved					
Attempt #	1	2	3	4	5
Date					
Score					

Claim:
There are different aquatic systems with diverse organisms.

Evidence	Reasoning/Justification
Aquatic ecosystems have abiotic - temp, light, salinity, depth	Different organisms can tolerate different factors
Some organisms prefer fresh water vs. saltwater. Bull sharks + manatees tolerate both	Bull sharks and manatees are very different, but tolerate same conditions
Depth reduces light and increases pressure. Organisms deal with this differently.	Many types of adaptations exist to survive in these conditions
The geography of an aquatic environment can determine the life forms present.	Continental shelf gets lots of light, so there's many living things. Not so in the abyss.
Photosynthesis takes place in the neritic zone.	It is the most productive area of the ocean. Plankton are base of food chain.
Metabolism is affected by temp. Many species are cold-blooded.	Organisms can migrate or must have adaptations to survive.

Note Taking Guide 4answers

Thomas Allen Reed



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web jul 27 2023 by identifying the limiting reactant calculating the amount of product formed and determining the excess reactant chemists can optimize reaction conditions and ensure efficient use of resources in this article we have explored the basics of limiting and excess reactants providing a comprehensive analysis of the pogil answer key for this

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web limiting and excess reactants 5 13 fill in the table below with the maximum moles of water that can be produced in each container q u indicate which reactant limits the quantity of water produced this is the limiting reactant also show how much of the other reactant the reactant in excess will be left over

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web the limiting reagent concept allows us to calculate amounts of reactants used and products formed in a complete chemical reaction based on the stoichiometric relationships in a balanced chemical equation so far we have considered these relationships on the basis of masses and moles

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web sep 24 2023 the pogil limiting and excess reactants answer key serves as a valuable resource for students to check their work and verify the accuracy of their calculations the answer key provides step by step explanations and calculations for determining the limiting reactant and the amount of excess reactant in a chemical reaction

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web 33mol o₂ 6mol br₂ 11mol o₂ 18mol br₂ identify the limiting reactant s and excess reactant s the limiting reactant is o₂ since it would yield the least amount of product 18 mol br₂ the excess reactant is c₂h₃br₃ since its complete reaction would have yielded up to 54 mol br₂ think about your result

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web step 4 the reactant that produces a smaller amount of product is the limiting reagent mg mg produces less mgo mgo than does o₂ o₂ 3 98 g mgo vs 25 2 g mgo therefore mg is the limiting reagent in this reaction step 5 the reactant that produces a larger amount of product is the excess reagent

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web please save your changes before editing any questions 5 minutes 1 pt viewing model 3 if 4 moles of hydrogen react with 2 moles of oxygen which of these is the excess reactant is leftover at the end and how many moles of it are leftover hydrogen 2 moles leftover hydrogen 1 mole leftover

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web if you start with 14 82 g of Ca(OH)_2 and 16 35 g of H_2SO_4 a determine the limiting reagent b determine the number of moles of H_2O produced c determine the number of grams of CaSO_4 produced d determine the number of grams of excess reagent left 1 make sure the equation is balanced

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