

Background

- 1. Study of this _____ multi _____ relationship is a _____ method _____
- 2. Research on _____ followed _____ approaches _____

2. $40g = 40r + 4 \cdot 40g$

2. Magnesium nitrate reacts with a single oxygen atom to produce 2 formula units of magnesium oxide.

The _____ is a _____ reaction _____ give the _____ _____ _____ from the _____ reaction in the _____

3. Problem

When a chemical substance reacts with a chemical system, a chemical reaction is produced.

$$2 \cdot A_2 + 3g \cdot B \rightarrow 2AB_2$$

moles of reactants

$$\begin{array}{cccc} 2 & A_2 & 3 & B \\ 2 & A_2 & 3 & AB_2 \\ 2 & B_2 & 3 & AB_2 \end{array}$$

2. If you start with 4 moles of A_2 , how many moles of AB_2 will be produced?

$$2 \cdot 4 \cdot 2 = 16 \text{ moles } AB_2 \text{ and } 3 \cdot 4 = 12 \text{ moles } B_2$$

Note Taking Guide Episode 801 Stoichiometry Answers

Judd E. Hollander



Note Taking Guide Episode 801 Stoichiometry Answers:

Adopting the Track of Term: An Psychological Symphony within **Note Taking Guide Episode 801 Stoichiometry Answers**

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