

### Chemical Formula

Talking about proportions, molecules and relative of atoms present in a compound.

Molecular  $\text{C}_{10}\text{H}_{12}\text{O}_4$  (molecular)

Formula represents the number of \_\_\_\_\_ of each  
in a single molecule.

Empirical  $\text{C}_5\text{H}_6\text{O}_2$  (molecular)

Formula represents the simplest ratio of  
positive and negative ions in a formula  
unit.

### Atomic Mass

The average mass of atoms present in a sample  
unit.

### Atomic

$^{12}\text{C}$  mass 12 atomic mass

$^{16}\text{O}$  mass 16 atomic mass  
unit

### Molecular

$\text{C}_{10}\text{H}_{12}\text{O}_4$  atomic mass

$\text{C}_{10}\text{H}_{12}\text{O}_4$  mass 208 atomic mass

$\text{C}_5\text{H}_6\text{O}_2$  atomic mass

# Note Taking Guide Episode 601 Chemical Formula Answers

**M Mark**



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