

Now, balance the first reaction the REDOX way! Underline the reducing agent and Circle the oxidizing agent.



### Learning Check

Why can't a REDOX reaction be balanced the normal way?



In a REDOX reaction the number of atoms and the number of electrons transferred needs to be balanced. In a normal reaction, only the number of atoms needs to be balanced. There is no transfer of electrons. As a result, the normal way of balancing won't work for a REDOX reaction. The electrons aren't being balanced.

# Note Packet 23 Oxidation Reduction Answer Key

**JS Bruner**



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