Polyatomic Ions Worksheet

Polyatomic Practice

KIRKITING PT-SOURCE		
Name or write the formula for the following	ng polyatomic ions	
suitate	CO ₃ ²	
nitribe	MnO ₃ ¹	
perphosphate	50 ₁ °	
hypolodite	BrO ₂ ¹	
chiloritie	CO ₄ P	
phosphile	PO ₃ 5	
percarbonate	CIO*	
bromate	1O2"	
hyposuffile	POv ^b	
permanganate	NO ₂ *	
carbonite	50v ⁰	
Name or write the formula for the following	ng Type I polyatomic ionic compounds	
beryllium hydroxide	BackOply	
sodium nitrite	Ga(GNO)s	
ammonium chloride	AgrSO ₄	
paloium bisulfate	MyCO	
rubidium perchionate	NH ₂ NO ₂	
strontium suffite	Alt(C ₀ H ₂ O ₀) ₀	
aluminum acetate	SrSO ₆	
ammonium nitrate	FISCIO ₂	
magnesium hypocarbonite	Ca(HSO _{id})	
silver hyposulfite	NHiGI	
gallium cyanate	NaNO	
barium lodate	Be(OH) ₃	
	Name or write the formula for the following suitable intribe perphosphate in hyposodital of fibritial phosphite percarbonate in hyposuffite permanganate carbonate in hyposuffite permanganate carbonate. Name or write the formula for the following beryffium hydroxide andium nitrite ammonium chloride calcium bisulfate nubidium perchlorate attrantium acetate ammonium nitrate magnesium hypocarbonite silver hyposuffite gallium cyanate.	Name or write the formula for the following polyatomic ions surface COs2 COs2

Polyatomic Ions Extension Questions Answers

Rachel S Tattersall

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