

## Part 2

# GCSE Chemistry OCR A (Combined Science - **HIGHER**) Paper 2 Exam Walkthrough

Combined Science Higher - Chemistry

Walkthrough Mark Paper

12  
19

Look at the equation for the equilibrium reaction between sulfur dioxide,  $\text{SO}_2$ , oxygen and sulfur trioxide,  $\text{SO}_3$ , in a closed system.



- (a) Predict the effect of adding more oxygen to the equilibrium mixture.

Explain your answer.

(2)

- (b) Predict the effect of increasing the total pressure in the equilibrium mixture.

Explain your answer.

(2)

- (c) Predict the effect of increasing the **temperature** of the equilibrium mixture.

Explain your answer.

(2)

- (d) Sulfur dioxide for this reaction is made by burning sulfur.



Calculate the mass of sulfur needed to make 48 tonnes of sulfur dioxide.

Give your answer to 2 significant figures.

The relative atomic mass,  $A_r$ , of S is 32.1 and of O is 16.0.

# Ocr Chemistry B 2013 B74

**Camilla Rothe**



**Ocr Chemistry B 2013 B74:**

## Unveiling the Energy of Verbal Artistry: An Emotional Sojourn through **Ocr Chemistry B 2013 B74**

In some sort of inundated with displays and the cacophony of instantaneous connection, the profound power and emotional resonance of verbal beauty frequently fade in to obscurity, eclipsed by the regular barrage of noise and distractions. Yet, set within the musical pages of **Ocr Chemistry B 2013 B74**, a captivating function of fictional elegance that pulses with raw thoughts, lies an memorable journey waiting to be embarked upon. Composed by a virtuoso wordsmith, that enchanting opus instructions viewers on a mental odyssey, softly exposing the latent potential and profound influence stuck within the intricate web of language. Within the heart-wrenching expanse of the evocative analysis, we can embark upon an introspective exploration of the book is key themes, dissect its charming publishing fashion, and immerse ourselves in the indelible impression it leaves upon the depths of readers souls.

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