

# PINKHASSOV

UN NOUVEAU REGARD SUR LA MOBILITÉ URBAINE

# Nouveau Regard Sur La Sociéteacute Franccedilaise

**Yan Bai**



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$$2\text{C}_6\text{H}_{10} + 17\text{O}_2 \rightarrow 12\text{CO}_2 + 10\text{H}_2\text{O}$$

$$\text{H}_2\text{O} + \text{KBr} + \text{CO}_2 \rightarrow \text{H}_2\text{SO}_4 + 2\text{H}_2\text{O} + \text{Na}_2\text{SO}_4$$
 using the following equation

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web may 28 2020 write the balanced equation for the reaction and calculate how much  $\text{SiO}_2$  is required to produce 3.00 kg of  $\text{SiC}$   

$$\text{SiO}_2 + 3\text{C} \rightarrow \text{SiC} + 2\text{CO}$$

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 $3\text{ HBr} + 1\text{ KHCO}_3 \rightarrow 1\text{ H}_2\text{O} + 1\text{ KBr} + \text{CO}_2$   
 $4\text{ GaBr}_3 + 3\text{ Na}_2\text{SO}_3 \rightarrow \text{Ga}_2\text{SO}_3 + 3\text{ NaBr} + 5\text{ SO}_2 + 3\text{ H}_2\text{O}$

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