ALKENES:

beneral formula = Cn.Hzm.

functional group: c=C(saturated)

REACTIONS

Hydration (420) => +-c-c-+

- outsides contrast - migh pressurer temp. - Forms excond

Halogenarion(Brz) => H

Hydrogenation(42) =7

- michiga coaronyst - Tempilso" - forms a licane

compustion = 7 complete: (02th20 incomplete: (0th20

WITCOINO IT:

functional group OH

General formula = Callenn OH

REACTIONS

Product: (0; +420 Burning =>

heaverouses. Add water (470) => (no H+ or OH-)

Add oxygen (02) = 7 forms (armoxytic audidi

, sugar t) yeast 3) mildion formentation => Ethanoll +) Time . . filtered remove yeast disn't remove water.

CABOXYLIC ACIDS:

functional group = COOH

General formula: (mHznerCOOH.

REACTIONS

Acid+Water=> (H3(00H= (H3(00+H+ weak acid-partially lonised w

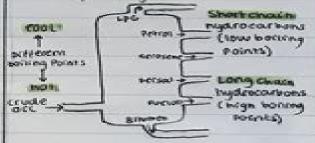
(Sandillaren (O.Decinos IV)) Acid+ (asbonate => (B3(00H+ Nas(0) -> (Dg+H10+CH3(00Nh (aq) due to CO)

ACCO + ALCOHOL -> CH3COOH + CH3CH2OH -> CH3CCOCH3CH2 + TH2O

eq emanoic acid+emanol > emyl emanoate.

FRACTIONAL DOSTULATION:

used to apperous mustocarbon fractions.



POLYMERISATION:

one monomer Addition Polymentsation

Condensation polymerisation . water side product

)-c-0H + HP

PO MOCO . 40- C-O-C-OH + H-N-D-N-H (N-N-D-NH)

- 12-0-8-M-10-M-1 +MEO

CRACKING

The breaking up the hydrocarbon chains which are too long from fractional distribution into Shorter More obeyout hydrocarbons.

TWO Types

O steam cracking: Vaporised hydrocarbon't steam 3-08 (Cataly increasing: Varioused by aracaroon + catalyst.

Example:

Calle HGHIN Calling =7

One alkene one alkane.

(7 H14+(H4)

6 HIZT CZHA

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S Ben Porath

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- 1. Understanding the eBook Notes On Organic Chemistry Chemistry 3 Klb
 - The Rise of Digital Reading Notes On Organic Chemistry Chemistry 3 Klb
 - Advantages of eBooks Over Traditional Books
- 2. Identifying Notes On Organic Chemistry Chemistry 3 Klb
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
- 3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - o Features to Look for in an Notes On Organic Chemistry Chemistry 3 Klb
 - User-Friendly Interface
- 4. Exploring eBook Recommendations from Notes On Organic Chemistry Chemistry 3 Klb
 - Personalized Recommendations
 - $\circ\,$ Notes On Organic Chemistry Chemistry 3 Klb User Reviews and Ratings
 - Notes On Organic Chemistry Chemistry 3 Klb and Bestseller Lists
- 5. Accessing Notes On Organic Chemistry Chemistry 3 Klb Free and Paid eBooks
 - Notes On Organic Chemistry Chemistry 3 Klb Public Domain eBooks
 - Notes On Organic Chemistry Chemistry 3 Klb eBook Subscription Services
 - Notes On Organic Chemistry Chemistry 3 Klb Budget-Friendly Options

- 6. Navigating Notes On Organic Chemistry Chemistry 3 Klb eBook Formats
 - o ePub, PDF, MOBI, and More
 - Notes On Organic Chemistry Chemistry 3 Klb Compatibility with Devices
 - Notes On Organic Chemistry Chemistry 3 Klb Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - o Adjustable Fonts and Text Sizes of Notes On Organic Chemistry Chemistry 3 Klb
 - Highlighting and Note-Taking Notes On Organic Chemistry Chemistry 3 Klb
 - Interactive Elements Notes On Organic Chemistry Chemistry 3 Klb
- 8. Staying Engaged with Notes On Organic Chemistry Chemistry 3 Klb
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Notes On Organic Chemistry Chemistry 3 Klb
- 9. Balancing eBooks and Physical Books Notes On Organic Chemistry Chemistry 3 Klb
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Notes On Organic Chemistry Chemistry 3 Klb
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Notes On Organic Chemistry Chemistry 3 Klb
 - Setting Reading Goals Notes On Organic Chemistry Chemistry 3 Klb
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Notes On Organic Chemistry Chemistry 3 Klb
 - Fact-Checking eBook Content of Notes On Organic Chemistry Chemistry 3 Klb
 - Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
- 14. Embracing eBook Trends
 - Integration of Multimedia Elements

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