

Ex. Phosphorus:

Students needed results with oxygen for products noted on their notes. How many grams of sodium would react to produce _____ g of sodium oxide?



4 2g Na₂O 1m Na₂O 4m Na 23Na
 62gNa₂O 3m Na₂O 1m Na
 31.16g of Na

When _____ g of hydrogen reacts with oxygen, how many grams of water are produced? $2\text{H}_2 + \text{O}_2 = 2\text{H}_2\text{O}$

12g H 1m H 2m H₂O 18g H₂O
 2g H 2m H 1m H₂O
 108g of H₂O

Actual Yield amount of product produced when the reaction _____
 is performed in a _____ lab

Theoretical Yield amount of product expected to be produced

based on the _____ balanced equation
 and the amount of _____ reactant

Percent Yield

Actual _____ yield / Theoretical _____ yield x 100%

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WJ Hussar



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