

## Extra Testing Section - (Experiments 1,2,3) - Part II

### Evidence for wave nature of light:

1. Interference experiments have revealed that light exhibits wave-like behavior in situations:
  - a. double-slit interference
  - b. thin film interference
  - c. diffraction
2. Huygens' principle can explain these patterns. Huygens is formulated as: a point on a wave behaves as if it were a source of spherical waves.
3. Huygens' principle, however, cannot provide a law to the behavior of light. The solution was to be the introduction of an electromagnetic wave equation that can predict wave behavior mathematically.

### Evidence for particle nature of light:

1. photoelectric effect: light propagates as waves
  - a. without carrying frequencies of light, which are characteristic of waves, electrons are ejected.
  - b. threshold frequency below which no electrons are ejected and above it electrons are ejected.
  - c. stopping potential is such as to cancel the kinetic energy of electrons that are ejected and above it electrons are ejected.
  - d. quantum energy is such as to cancel the kinetic energy of electrons that are ejected and above it electrons are ejected.

<p>Student should fill light <u>propagates</u> <u>as waves</u> and <u>cancel</u> <u>the kinetic energy</u> of electrons that are ejected and above it electrons are ejected.</p>
<p>Without this photoelectric effect, it would be <u>impossible</u> <u>to</u> <u>cancel</u> <u>the kinetic energy</u> of electrons that are ejected and above it electrons are ejected.</p>
<p>Without light <u>propagates</u> <u>as waves</u> <u>cancel</u> <u>the kinetic energy</u> of electrons that are ejected and above it electrons are ejected.</p>

2. The wave theory cannot explain the photoelectric effect

# Note Taking Guide Episode 1302 Part 1 Answers

**L Cohen**



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