Lab POGIL: Metal Reactivity - Introduction & Pre-Lab

Discovery Question: How can trends in metal reactivity be used to predict the products of chemical reactions?

Objectives:

- Develop a procedure to test the reactivity of metals.
- List metals in order of reactivity based on the results of your procedure
- Predict the products of chemical reactions by considering the relative reactivity of the reactants

You have been provided with vials containing 3 types of metals (A. B. C). You also have vials tilled with solutions containing cations of each metal (A", B", C"), and an additional vial containing a solution of cation D". You will also have (6-well x 4-well) plates available for small-scale reactions of metals and ionic solutions.

| Elemental | Solutions | | |
|-------------|---|--|--|
| Metals | Containing Metal Cations | | |
| A B C | A (contains A ^{**}) B (contains B ^{**}) C (contains C ^{**}) D (contains D ^{**}) | | |

Remember to wear goggles at all times. Avoid skin contact with all solutions. If contact occurs, rinse with water immediately. Solution D will stain skin and clothes.

Pre-lab Requirements:

 Describe a method for predicting charges of ions formed from single atoms of an element using the periodic table.

Use a periodic table to predict charges for ions formed from each element listed below:

| Element lo | 5OL | Element | Ion | Element | Ion |
|------------------|-----|----------------|------|---------------|-----|
| Lithium (Li) | | Beryllium (Be) | | Copper (Cu) | _/ |
| Sodium (Na) | | Magnesium (Mg) | | Zimc (Zn) | |
| Potassium (K) | | Calcium (Ca) | | Silver (Ag) | |
| Nitrogen (N) | - | Oxygen (O) | | Pluorine (F) | |
| Phosphorous (P)_ | | Sulfur (S) | 2222 | Chlorine (CI) | |
| | | Selenium (Se) | | Bromine (Br) | |

Reactivity Of Metals Pogil

Krystyna Dyrek, Jerzy Haber, Janusz Nowotny **Reactivity Of Metals Pogil:**

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