Identifying Reaction Types & Balancing Equations

Step 1: Name the type of chemical reaction	Step 2: Balance the following chemical equations:
Ex. Combination Reaction	2_N₂ + 6_H₂ □ 4_NH₃
1.	KCIO, KCI + O ₂
2.	NaCl +F₂ □NaF + Cl₂
3.	H ₂ + O ₂ H ₂ O
4.	AgNO ₃ +MgCl ₂
5.	AlBr ₃ +K ₂ SO ₄
6.	CH ₄ +O ₂ □CO ₂ +H ₂ O
7.	C2H2 +O2
8.	FeCl ₃ +NaOH 🗆Fe(OH) ₃ +NaCl
9.	P + O,
10.	Na + H ₂ O □NaOH + H ₂
11.	Ag ₂ OAg + O ₂
12.	CO ₂ +H ₂ O □C ₆ H ₁₂ O ₆ + O ₂
13.	K +MgBr₂ □ KBr +Mg
14.	HNO ₂ (aq) +Ba(OH) ₂ (aq) Ba(NO ₂) ₂ (aq) +H ₂ O (I)
15.	C _i H _{1:} (g) + O _i (g) □ CO _i (g) + H _i O(g)
16.	Al(s) + Fe ₂ O ₃ (aq) Al ₂ O ₃ (aq) + Fe(s)
17.	Al(s) + O ₂ (g)
18.	P ₄ + O ₂ □ 2 P ₂ O ₃
19.	_2_CaCO;
20.	H ₂ + O ₂ □ 2H ₂ O

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