## **Predicting Reaction Products**

Predict the products for the following reactions, balance the equation, then classify the type of reaction:

- Na + FeBr<sub>3</sub> → 1)
- 2)  $NaOH + ___ H_2SO_4 \rightarrow$
- 3)  $C_2H_4O_2 + O_2 \rightarrow$
- $NH_1 + H_2O \rightarrow$ 40
- 5): PbSO<sub>4</sub> + \_\_\_AgNO<sub>3</sub> →
- 6)  $PBr_s \rightarrow$
- 7) HBr + Fe →
- KMnO<sub>4</sub> + ZnCl<sub>2</sub> → 8)
- 90  $MnO_2 + Sn(OH)_4 \rightarrow$
- $O_2 + C_3H_{12}O_2 \Rightarrow$ 10)
- 11) H<sub>2</sub>O<sub>2</sub> →
- 12) PtCl<sub>4</sub>+ Cl<sub>2</sub> →

# **Reaction Prediction Practice Answer Key**

**Neil E. Schore** 

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