

Reaction Energy

Question 1: Reaction Energy

In the space provided, describe the term as it is used in physics that best completes each sentence or argument below.

1. The sign convention for the study of thermodynamics is that
 - a. energy = energy being added.
 - b. energy = energy being + heat + work.
 - c. work = energy being thermal systems.
 - d. heat is work.
2. Energy is measured in units of
 - a. joules.
 - b. calories.
 - c. grams.
 - d. grams.
3. Heat added to a system is called _____ energy.
 - a. Q
 - b. W
 - c. U
 - d. None of the above.
4. The specific heat of a substance is the amount of energy needed to

raise the temperature of

 - a. 1 g by 1 °C.
 - b. 1 calorie 1 g.
 - c. 1 kg by 1 °C.
 - d. 1 calorie 1 °C.
5. Methods of the following to increase the temperature (the energy) supplied to

increase the temperature of a substance:

 - a. decrease of the substance.
 - b. the specific heat of the substance.
 - c. decrease mass of the substance.
 - d. decrease of heat.

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