



ANSWER SHEET

For each answer, please fill in marks like this

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RRB Group D Answer Key

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EW Minium



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negative when heat flows from the system to the surroundings thus exothermic reactions have a negative Δh of reaction the
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historically this reaction was represented as acid base alkali salt water for example $\text{hcl naoh nacl h}_2\text{o}$

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energy is released which is detected by the thermometer

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web for reactions involving acetic acid or ammonia the measured enthalpy change of neutralization is a few kJ less exothermic than with strong acids and bases for example one source which gives the enthalpy change of neutralization of sodium hydroxide solution with HCl as 57.9 kJ mol⁻¹

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