

B-3. Dissolution Rates: Practice Problems

1. Consider the reaction of the gasification of coal: (4) B. (100%)



It is assumed to proceed via a surface reaction, i.e. at a metal or its oxide. How would you estimate the results of a gas activation?

2. The unbalanced reaction for the steam-gasification of carbon is shown (20% weight) in the following reaction:



a. Write the rate law for the reaction of the carbon.

- b. The reaction is assumed to proceed via a surface reaction, i.e. at a metal or its oxide. How would you estimate the results of a gas activation?

c. The reaction is assumed to proceed via a surface reaction, i.e. at a metal or its oxide. How would you estimate the results of a gas activation?

3. The reaction is shown in the following reaction:

Reaction Rates And Equilibrium Practice Problems Answer

Michal Rosen-Zvi



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