

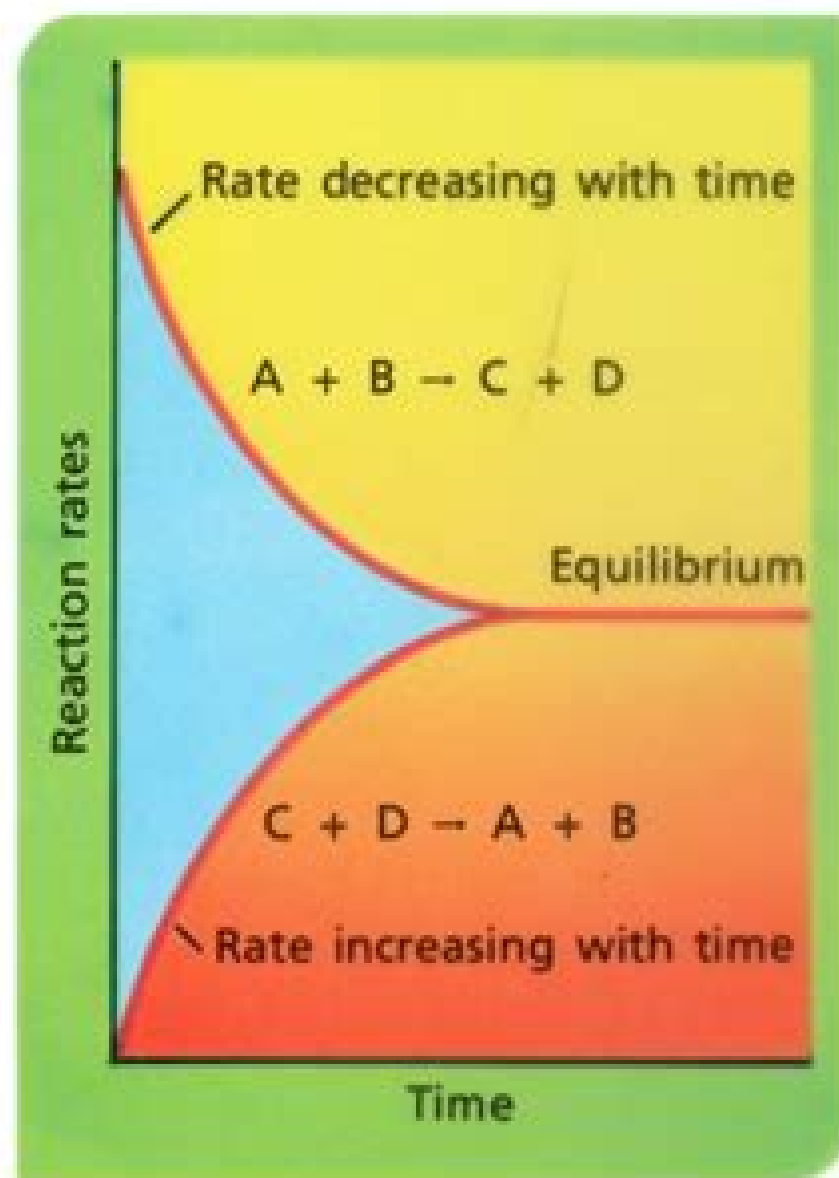
Rate vs Equilibrium

At Equilibrium: RATES ARE EQUAL

- the **concentrations** of reactants and products are **constant**.

🔧 $[]'s = 0$

- The forward and reverse reactions continue after equilibrium is attained.



Reaction Rates And Equilibrium

K Payea



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