

Worksheet 1-1

- Measuring Reaction Rates

 Name - _____
 Due Date - _____

Answer the questions in the space provided. You must show all of your work to receive full marks. All answers MUST be rounded to the correct number of Sig. Figures.

1. A chemist wishes to determine the rate of reaction of zinc with hydrochloric acid. The equation for the reaction is:



A piece of zinc is dropped into 1.00 L of 0.100 M HCl and the following data were obtained:

Time	Mass of Zinc
0 s	0.016 g
4 s	0.014 g
8 s	0.012 g
12 s	0.010 g
16 s	0.008 g
20 s	0.006 g

$$\begin{array}{r} 0.016\text{g} \\ - 0.006\text{g} \\ \hline 0.010\text{g} \end{array}$$

- a) Calculate the **Rate of Reaction** in grams of Zn consumed per second.

$$\text{rate} = \frac{\Delta g}{\Delta \text{time}} = \frac{0.010\text{g}}{20\text{sec}}$$

Answer 5.0×10^{-4}

- b) Calculate the **Rate of Reaction** in moles of Zn consumed per second.

$$5.0 \times 10^{-4} \frac{\text{g}}{\text{sec}} \times \frac{1\text{mol}}{65.4\text{g}} =$$

Answer $7.6 \times 10^{-6} \frac{\text{mol}}{\text{sec}}$

- c) Write out the complete ionic equation for the reaction.



- d) What will happen to the $[\text{H}^+]$ as the reaction proceeds? decrease

- e) What will happen to the $[\text{Cl}^-]$ as the reaction proceeds? nothing, it is a spectator

Reaction Rate Worksheet Answers

Daniela Niemeyer

A red circular graphic with a gradient, appearing as a semi-circle or a partial circle, located to the right of the author's name.

Reaction Rate Worksheet Answers:

Decoding **Reaction Rate Worksheet Answers**: Revealing the Captivating Potential of Verbal Expression

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Table of Contents Reaction Rate Worksheet Answers

1. Understanding the eBook Reaction Rate Worksheet Answers
 - The Rise of Digital Reading Reaction Rate Worksheet Answers
 - Advantages of eBooks Over Traditional Books
2. Identifying Reaction Rate Worksheet Answers
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Reaction Rate Worksheet Answers
 - User-Friendly Interface
4. Exploring eBook Recommendations from Reaction Rate Worksheet Answers
 - Personalized Recommendations
 - Reaction Rate Worksheet Answers User Reviews and Ratings
 - Reaction Rate Worksheet Answers and Bestseller Lists

5. Accessing Reaction Rate Worksheet Answers Free and Paid eBooks
 - Reaction Rate Worksheet Answers Public Domain eBooks
 - Reaction Rate Worksheet Answers eBook Subscription Services
 - Reaction Rate Worksheet Answers Budget-Friendly Options
6. Navigating Reaction Rate Worksheet Answers eBook Formats
 - ePub, PDF, MOBI, and More
 - Reaction Rate Worksheet Answers Compatibility with Devices
 - Reaction Rate Worksheet Answers Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Reaction Rate Worksheet Answers
 - Highlighting and Note-Taking Reaction Rate Worksheet Answers
 - Interactive Elements Reaction Rate Worksheet Answers
8. Staying Engaged with Reaction Rate Worksheet Answers
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Reaction Rate Worksheet Answers
9. Balancing eBooks and Physical Books Reaction Rate Worksheet Answers
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Reaction Rate Worksheet Answers
10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine Reaction Rate Worksheet Answers
 - Setting Reading Goals Reaction Rate Worksheet Answers
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Reaction Rate Worksheet Answers
 - Fact-Checking eBook Content of Reaction Rate Worksheet Answers
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning

- Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
14. Embracing eBook Trends
- Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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