Predicting Reaction Products

Predict the products for the following reactions, balance the equation, then classify the type of reaction:

- Na + ___ FeBr₃ →
- 2) ____ NaOH + ____ H₂SO₄ →
- 3) ____ C₂H₄O₂ + ____ O₂ →
- 4) ____ NH₃ + ____ H₂O →
- 5) ____ PbSO₄ + ____ AgNO₃ →
- 6) _____PBr₃ →
- 7) HBr + Fe →
- 8) ____ KMnO₄ + ___ ZnCl₂ →
- 9) ____MnO₂ + ____ Sn(OH)₄ →
- 10) ____ O₂ + ____ C₃H₁₂O₂ →
- 11) ____ H₂O₂ →
- 12) ____ PtCl₄ + ___ Cl₂ →

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Ingo Waldemar Dagobert Hackh

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Chemical Reactions and Their Equations Ingo Waldemar Dagobert Hackh, 1921 The State Identification of **Reaction Products** Richard N. Zare, STANFORD UNIV CA DEPT OF CHEMISTRY., 1982 We have employed lasers to study elementary chemical reactions and to examine the structure of molecular ions Specifically we have 1 investigated the effect of atomic reagent approach geometry on the outcome of chemical reaction 2 examined ion unimolecular fragmentation kinetics using a novel multiphoton ionization technique 3 studied the molecular CO2 ion using doubleresonance spectroscopy and 4 developed a new method for determining the vibrational structure of molecular ions Author Chemical Reactivity in Liquids Michael Moreau, Pierre Turg, 2012-12-06 Understanding chemical reactivity has been the permanent concern of chemists from time immemorial If we were able to understand it and express it quantitatively there would practically remain no unsolved mystery and reactions would be fully predictable with their products and rates and even side reactions The beautiful developments of thermodynamics through the 19th century supplied us with the knowledge of the way a reactions progresses and the statistical view initiated by Gibbs has progressively led to an unders tanding closer to the microscopic phenomena But is was always evident to all that these advances still left our understanding of chemical reactivity far behind our empirical knowledge of the chemical reaction in its practically infinite variety. The advances of recent years in quantum chemistry and statistical mechanics enhanced by the present availability of powerful and fast computers are very fast changing this picture and bringing us really close to a microscopic understanding of chemical equilibria reaction rates etc This is the reason why our Society encouraged a few years ago the initiative of Professor Savo Bratos who with a group of French colleagues prepared an impressive study on Reactivite chimique en phase liquide a prospective report which was jointly published by the Societe Fran **Consecutive Chemical Reactions** Nikolaĭ Mikhaĭlovich Rodigin,Ėmilii a Nikolaevna Rodigina, 1964 Multicomponent Reactions Jieping Zhu, Hugues Bienaymé, 2005-03-04 In the very first book on this hot topic the expert editors and authors present a comprehensive overview of these elegant reactions From the contents Organoboron compounds Free radical mediated multicomponent coupling reactions Applications in drug discovery Metal catalyzed reactions Total synthesis of natural products Asymmetric isocyanide based reactions The Biginelli reaction Asymmetric isocyanide based reactions The Domino Knoevenagel Hetero Diels Alder Reaction and related transformations Catalytic asymmetric reactions Algorithm based methods for discovering novel reactions Post condensation modifications of the Passerini and Ugi reactions An essential reference for organic and catalytic chemists and those working in organometallics both in academia and industry

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