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web log $[\text{H}^+]$ 10⁻⁴ 0.5891 x 10⁻⁵ mol dm⁻³ log pK_a 3.75 4.05 0.30 10.0 3.0 0.50 and HCOOH 0.1 so HCOO^- 0.200 mol dm⁻³ amount vol x

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