

Types of Chemical Reactions: Study Hall Chemistry #2

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i. States of matter describe the _____ a substance can take.
This can be a _____, or an
_____ solution, which is dissolved in water.

a. States of matter clarify what _____ the
substance has _____ the reaction.

ii. A _____, or **synthesis**
reaction, is the merging of two or more substances
to create one product.

iii. A _____ is the separation of a combined
substance into multiple products.

1. A _____ is a substance that facilitates a reaction.

2. What is a **single displacement**? Explain.

a. Likewise, a _____ reaction follows
the same pattern with two substances.

3. How do **cations** differ from **anions**? Explain.

iii. A _____ can produce a gas or a liquid and a gas.

iv. _____ are also defined by their products,
this time when one's a solid.

v. _____ is the gaining of an electron. _____ is
the loss of an electron.

vi. _____ typically involve _____
which are molecules made up of hydrogen and carbon.

1. Combustion is _____, meaning it gives off heat during
the reaction.

vii. How does an **acid** differ from a **base**?



anion



cation

Patience Hall Chemistry 6 3 Worksheet

Mark Steyvers



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