



## **Class 10 Physical Science**

### **Chemistry Formal Experiment: *Using flame colours to identify unknown metal ions***

**August 2024**

**60 Marks**

**1 hour**

**Full Name:** .....

#### **Introduction:**

The flame test is used to visually determine the identity of an unknown metal or metalloid, as an element or in a compound, based on the characteristic colour the salt turns the flame of a burner. When an electron in an excited state (higher energy) returns to its ground state (lower energy), an atomic spectrum is produced. A specific quantity of energy is associated with each electron, and when an electron returns to a lower energy level, a specific quantity of energy is released. The energy is released as electromagnetic radiation. Certain wavelengths will be visible to the naked eye; other wavelengths can only be detected with instruments. In the activity below, we will investigate the colours emitted by certain metal atoms. Different colours indicate different wavelengths of visible radiation.

#### **Materials:**

- ☐  $\text{NaCl(aq)}$ ,  $\text{KCl(aq)}$ ,  $\text{CaCl}_2 \cdot 2\text{H}_2\text{O(aq)}$ ,  $\text{CuCl}_2 \cdot 2\text{H}_2\text{O (aq)}$ ,
- ☐ copper powder, magnesium powder,  
zinc powder, iron powder
- ☐ four unknown salt solutions
- ☐ Bunsen burner
- ☐ Nichrome wire



# Physical Science Formal Experiment Nomemo 2014

**LP Steffe**



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## **Table of Contents Physical Science Formal Experiment Nomemo 2014**

1. Understanding the eBook Physical Science Formal Experiment Nomemo 2014
  - The Rise of Digital Reading Physical Science Formal Experiment Nomemo 2014
  - Advantages of eBooks Over Traditional Books
2. Identifying Physical Science Formal Experiment Nomemo 2014
  - Exploring Different Genres
  - Considering Fiction vs. Non-Fiction
  - Determining Your Reading Goals
3. Choosing the Right eBook Platform
  - Popular eBook Platforms
  - Features to Look for in an Physical Science Formal Experiment Nomemo 2014
  - User-Friendly Interface
4. Exploring eBook Recommendations from Physical Science Formal Experiment Nomemo 2014
  - Personalized Recommendations
  - Physical Science Formal Experiment Nomemo 2014 User Reviews and Ratings
  - Physical Science Formal Experiment Nomemo 2014 and Bestseller Lists
5. Accessing Physical Science Formal Experiment Nomemo 2014 Free and Paid eBooks
  - Physical Science Formal Experiment Nomemo 2014 Public Domain eBooks
  - Physical Science Formal Experiment Nomemo 2014 eBook Subscription Services
  - Physical Science Formal Experiment Nomemo 2014 Budget-Friendly Options

6. Navigating Physical Science Formal Experiment Nomemo 2014 eBook Formats
  - ePub, PDF, MOBI, and More
  - Physical Science Formal Experiment Nomemo 2014 Compatibility with Devices
  - Physical Science Formal Experiment Nomemo 2014 Enhanced eBook Features
7. Enhancing Your Reading Experience
  - Adjustable Fonts and Text Sizes of Physical Science Formal Experiment Nomemo 2014
  - Highlighting and Note-Taking Physical Science Formal Experiment Nomemo 2014
  - Interactive Elements Physical Science Formal Experiment Nomemo 2014
8. Staying Engaged with Physical Science Formal Experiment Nomemo 2014
  - Joining Online Reading Communities
  - Participating in Virtual Book Clubs
  - Following Authors and Publishers Physical Science Formal Experiment Nomemo 2014
9. Balancing eBooks and Physical Books Physical Science Formal Experiment Nomemo 2014
  - Benefits of a Digital Library
  - Creating a Diverse Reading Collection Physical Science Formal Experiment Nomemo 2014
10. Overcoming Reading Challenges
  - Dealing with Digital Eye Strain
  - Minimizing Distractions
  - Managing Screen Time
11. Cultivating a Reading Routine Physical Science Formal Experiment Nomemo 2014
  - Setting Reading Goals Physical Science Formal Experiment Nomemo 2014
  - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Physical Science Formal Experiment Nomemo 2014
  - Fact-Checking eBook Content of Physical Science Formal Experiment Nomemo 2014
  - Distinguishing Credible Sources
13. Promoting Lifelong Learning
  - Utilizing eBooks for Skill Development
  - Exploring Educational eBooks
14. Embracing eBook Trends
  - Integration of Multimedia Elements

- Interactive and Gamified eBooks

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