

Physical Science
Chapter 2

Properties of Matter

Classifying Matter

Substances

- **Element:** a substance that can't be separated down into simpler substances.
- **Atom:** the smallest particle of an element.
- An element has a fixed composition because it contains only one type of atom.

2 Types of Mixtures

1. **Heterogeneous Mixture:** have materials that are not evenly distributed and noticeably different from one another.
 2. **Homogeneous Mixture:** have materials that are evenly distributed so it is difficult to distinguish one substance from another.
 - a. Also called a solution
 - b. Remain constantly and uniformly mixed
- Which type of mixture is milk?
 - Based on the size of its largest particles, a mixture can be classified as a solution, a suspension, or a colloid.

Solutions

Solution: when substances dissolve and form a homogenous mixture.

- Examples: windshield wiper fluid, sea water, lemonade

Colloids

Colloid: contains some particles that are larger than those in a solution, but smaller than those in a suspension.

- Examples: homogenized milk, fog
- Think about it...
 - What happens to the size of the drops of fat in milk when milk is homogenized?

The Scattering of Light

- Colloids and Suspensions scatter light just like fog.
- Light goes right through a solution.

Chemical Properties

- Any ability to produce a change in the composition of matter
- Chemical properties can be observed only when the substance in a sample of matter is changing into different substances.

Physical Science Chapter 2 Review Answers Properties Of Matter

**Barron's Educational Series, Mark
Kernion, Joseph A. Mascetta**



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Holt Physical Science Mapi M. Cuevas, 1994

Resources for Teaching Middle School Science Smithsonian Institution, National Academy of Engineering, National Science Resources Center of the National Academy of Sciences, Institute of Medicine, 1998-04-30 With age appropriate inquiry centered curriculum materials and sound teaching practices middle school science can capture the interest and energy of adolescent students and expand their understanding of the world around them Resources for Teaching Middle School Science developed by the National Science Resources Center NSRC is a valuable tool for identifying and selecting effective science curriculum materials that will engage students in grades 6 through 8 The volume describes more than 400 curriculum titles that are aligned with the National Science Education Standards This completely new guide follows on the success of Resources for Teaching Elementary School Science the first in the NSRC series of annotated guides to hands on inquiry centered curriculum materials and other resources for science teachers The curriculum materials in the new guide are grouped in five chapters by scientific area Physical Science Life Science Environmental Science Earth and Space Science and Multidisciplinary and Applied Science They are also grouped by type core materials supplementary units and science activity books Each annotation of curriculum material includes a recommended grade level a description of the activities involved and of what students can be expected to learn a list of accompanying materials a reading level and ordering information The curriculum materials included in this book were selected by panels of teachers and scientists using

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Whitman-Peck Physics Walter George Whitman,1946 *Fundamentals of Electrochemical Science* Keith Oldham,Jan Myland,2012-12-02 Fundamentals of Electrochemical Science is a valuable contribution and I support the publication I am looking forward to seeing this book on the shelves and once published I will not hesitate to recommend itto my students ANDRZEJ WIECKOWSKI University of Illinois at Urbana Champaign Deals comprehensively with the basic science of electrochemistry Treats electrochemistry as a discipline in its own right and not as a branch of physical or analytical chemistry Provides a thorough and quantitative description of electrochemical fundamentals

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