

# Half Life Calculations Practice

2 pages + 2-page Answer Key

## HALF-LIFE CALCULATIONS

1. Radioiodine has a half-life of 8 days. If you start with 40 grams of radioiodine, how much will remain after 16 days?
2. If the half-life of Carbon-14 is 5730 years, how much of a 100 gram sample will remain after 11460 years?
3. If the half-life of Uranium-238 is 4.5 billion years, how long will it take for Uranium-238 to decay to 1/16 of the original amount?
4. The half-life of Uranium-238 is 4.5 billion years. How long will it take for a sample of Uranium-238 to decay to 1/16 of the original amount?
5. Radium-226 has a half-life of 1600 years. If you start with 64 grams, how much will remain after 4800 years?
6. If the half-life of Potassium-40 is 1.25 billion years, how long will it take for Potassium-40 to decay to 1/16 of the original amount?

## HALF-LIFE CALCULATIONS

1. Radioiodine has a half-life of 8 days. If you start with 40 grams of radioiodine, how much will remain after 16 days?  
 $40 \rightarrow 20 \rightarrow 10$   
10g will remain after 16 days
2. If the half-life of Carbon-14 is 5730 years, how much of a 100 gram sample will remain after 11460 years?  
 $100 \rightarrow 50 \rightarrow 25$   
25g will remain after 11460 years
3. If the half-life of Uranium-238 is 4.5 billion years, how long will it take for Uranium-238 to decay to 1/16 of the original amount?  
 $1 \rightarrow 1/2 \rightarrow 1/4 \rightarrow 1/8 \rightarrow 1/16$ , so 4 half-lives  
4.5 billion years  $\times$  4 half-lives  
= 18 billion years will remain
4. The half-life of Uranium-238 is 4.5 billion years. How long will it take for a sample of Uranium-238 to decay to 1/16 of the original amount?  
 $100 \rightarrow 50 \rightarrow 25 \rightarrow 12.5 \rightarrow 6.25$   
6.25g will remain after 18 billion years
5. Radium-226 has a half-life of 1600 years. If you start with 64 grams, how much will remain after 4800 years?  
 $64 \rightarrow 32 \rightarrow 16 \rightarrow 8$   
8g will remain after 4800 years
6. If the half-life of Potassium-40 is 1.25 billion years, how long will it take for Potassium-40 to decay to 1/16 of the original amount?  
 $1 \rightarrow 1/2 \rightarrow 1/4 \rightarrow 1/8 \rightarrow 1/16$ , so 4 half-lives  
1.25 billion years  $\times$  4 half-lives  
= 5 billion years will remain

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# Physical Science Half Life Calculations Instructional Fair Inc

**Yijin Wang**



## **Physical Science Half Life Calculations Instructional Fair Inc:**

## Decoding **Physical Science Half Life Calculations Instructional Fair Inc**: Revealing the Captivating Potential of Verbal Expression

In an era characterized by interconnectedness and an insatiable thirst for knowledge, the captivating potential of verbal expression has emerged as a formidable force. Its power to evoke sentiments, stimulate introspection, and incite profound transformations is genuinely awe-inspiring. Within the pages of "**Physical Science Half Life Calculations Instructional Fair Inc**," a mesmerizing literary creation penned with a celebrated wordsmith, readers attempt an enlightening odyssey, unraveling the intricate significance of language and its enduring effect on our lives. In this appraisal, we shall explore the book is central themes, evaluate its distinctive writing style, and gauge its pervasive influence on the hearts and minds of its readership.

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