#### Physical Science Chapter 2

#### Properties of Matter

#### Classifying Matter

Nucleotameen

- Element: a substance that can't be separated down into simpler substances.
- Atom: the smallest particle of an element.
- An element has a fixed composition because it contains only one type of atom.

2 Types of Mixtures

- Heterogeneous Mixture: have materials that are not evenly distributed and noticeably different from one another.
- Homogeneous Mixture: have materials that are evenly distributed so it is difficult to distinguish one substance from another.
  - a. Also called a solution
  - Remain constantly and uniformly mixed
- Which type of mixture is milk?
  - Based on the size of its largest particles, a mixture can be classified as a solution, a suspension, or a colloid.

Solutions

Solution: when substances dissolve and form a homogenous mixture.

Examples: windshield wiper fluid, sea water, lemonade

#### Colloids

Colloid: contains some particles that are larger than those in a solution, but smaller than those in a suspension.

- Examples: homogenized milk, fog
- Think about it...
  - o What happens to the size of the drops of fat in milk when milk is homogenized?

#### The Scattering of Light

- · Colloids and Suspensions scatter light just like fog.
- Light goes right through a solution.

#### Chemical Properties

- Any ability to produce a change in the composition of matter
- Chemical properties can be observed only when the substance in a sample of matter is changing into different substances.

**R Barnett** 

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