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Klausur: Physikalische Chemie

Physikalische Chemie: Chapter 10: Outer Thermal Energy and Work

Multiple Choice:

1. The heat reservoirs (heatbath) is characterized by:
- a) **infinite capacity**
 - b) large heat capacity
 - c) large mass
 - d) large size
2. Energy added to a system:
- a) **increases entropy, increases enthalpy**
 - b) reduces heat transfer
 - c) is irreversible
 - d) reduces the kinetic energy of the system
3. When a definite quantity heat is added to an object:
- a) temperature will be lower
 - b) **temperature is not sure to change**
 - c) there will be 100% efficiency
 - d) **it takes more time to get to final temp**
4. A heat engine transforms thermal energy mainly by:
- a) **compression**
 - b) expansion
 - c) specific heat
 - d) temperature
5. Compression of a substance is:
- a) **irreversible**
 - b) reversible
 - c) **disorderly change**
 - d) **spontaneous**
6. Why does liquid water is thermodynamically:
- a) **irreversible**
 - b) **irreversible**
 - c) **irreversible**
 - d) **irreversible**
7. Which statement is NOT true about heat?
- a) **it is a scalar quantity**
 - b) **it is a scalar quantity**
 - c) **it is a scalar quantity**
 - d) **it is a scalar quantity**
8. Which of the following is a thermal process?
- a) **irreversible**
 - b) **irreversible**
 - c) **irreversible**
 - d) **irreversible**

Physical Science Chapter 16 Test

G Psacharopoulos



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