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Your Roll No.....

5844

B.Sc. (Hons.)/III/NS  
CHEMISTRY—Paper XVII  
(Physical Chemistry—III)

F

Time : 3 Hours

Maximum Marks : 38

(Write your Roll No. on the top immediately on receipt of this question paper.)

Attempt six questions in all.

Q. No. 1 is compulsory. At least one question  
is to be attempted from each Section.

1. Explain briefly any *four* of the following :  $4 \times 2 = 8$
- (a) Maximum order of a reaction is restricted to three only.
  - (b) A photochemical reaction with quantum efficiency of one shows no temperature dependence.
  - (c) Adsorption of gases on the surface of adsorbent is not monolayer at high pressure and low temperature.
  - (d) Why can we not use a voltmeter for determining the electromotive force of a galvanic cell ?
  - (e) A first order reaction never completes.

P.T.O.

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**J Rink**



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## **Table of Contents Physical Question Paper Grade 1march 2014**

1. Understanding the eBook Physical Question Paper Grade 1march 2014
  - The Rise of Digital Reading Physical Question Paper Grade 1march 2014
  - Advantages of eBooks Over Traditional Books
2. Identifying Physical Question Paper Grade 1march 2014
  - Exploring Different Genres
  - Considering Fiction vs. Non-Fiction
  - Determining Your Reading Goals
3. Choosing the Right eBook Platform
  - Popular eBook Platforms
  - Features to Look for in an Physical Question Paper Grade 1march 2014
  - User-Friendly Interface
4. Exploring eBook Recommendations from Physical Question Paper Grade 1march 2014
  - Personalized Recommendations
  - Physical Question Paper Grade 1march 2014 User Reviews and Ratings
  - Physical Question Paper Grade 1march 2014 and Bestseller Lists
5. Accessing Physical Question Paper Grade 1march 2014 Free and Paid eBooks
  - Physical Question Paper Grade 1march 2014 Public Domain eBooks
  - Physical Question Paper Grade 1march 2014 eBook Subscription Services
  - Physical Question Paper Grade 1march 2014 Budget-Friendly Options

6. Navigating Physical Question Paper Grade 1march 2014 eBook Formats
  - ePub, PDF, MOBI, and More
  - Physical Question Paper Grade 1march 2014 Compatibility with Devices
  - Physical Question Paper Grade 1march 2014 Enhanced eBook Features
7. Enhancing Your Reading Experience
  - Adjustable Fonts and Text Sizes of Physical Question Paper Grade 1march 2014
  - Highlighting and Note-Taking Physical Question Paper Grade 1march 2014
  - Interactive Elements Physical Question Paper Grade 1march 2014
8. Staying Engaged with Physical Question Paper Grade 1march 2014
  - Joining Online Reading Communities
  - Participating in Virtual Book Clubs
  - Following Authors and Publishers Physical Question Paper Grade 1march 2014
9. Balancing eBooks and Physical Books Physical Question Paper Grade 1march 2014
  - Benefits of a Digital Library
  - Creating a Diverse Reading Collection Physical Question Paper Grade 1march 2014
10. Overcoming Reading Challenges
  - Dealing with Digital Eye Strain
  - Minimizing Distractions
  - Managing Screen Time
11. Cultivating a Reading Routine Physical Question Paper Grade 1march 2014
  - Setting Reading Goals Physical Question Paper Grade 1march 2014
  - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Physical Question Paper Grade 1march 2014
  - Fact-Checking eBook Content of Physical Question Paper Grade 1march 2014
  - Distinguishing Credible Sources
13. Promoting Lifelong Learning
  - Utilizing eBooks for Skill Development
  - Exploring Educational eBooks
14. Embracing eBook Trends
  - Integration of Multimedia Elements

- Interactive and Gamified eBooks

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