

# SECOND MID TERM TEST - 2024

Standard XII

Reg.No. 

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CHEMISTRY

Time : 1.30 hrs

Part - I

Marks :

10 x 1 =

1. Choose the correct answer:

- A magnetic moment of 1.73 BM will be shown by one among the following  
a)  $\text{TiCl}_4$                       b)  $[\text{CoCl}_6]^{4-}$                       c)  $[\text{Cu}(\text{NH}_3)_4]^{2+}$                       d)  $[\text{Ni}(\text{CN})_4]^{2-}$
- Which type of isomerism is exhibited by  $[\text{Pt}(\text{NH}_3)_2 \text{Cl}_2]$  ?  
a) coordination isomerism                      b) linkage isomerism  
c) optical isomerism                      d) geometrical isomerism
- The number of electrons that have a total charge of 9650 coulombs is  
a)  $6.22 \times 10^{23}$                       b)  $6.022 \times 10^{24}$                       c)  $6.022 \times 10^{22}$                       d)  $6.022 \times 10^{-34}$
- Which of the following electrolytic solution has the least specific conductance?  
a) 2 N                      b) 0.002 N                      c) 0.02 N                      d) 0.2 N

# Physical Chemistry Second Midtterm Exam Questions

**Robert R. Redfield**



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