

FINAL EXAMANITION IN PHYSICAL SCIENCE

I. Multiple Choices: For each of the following items, choose the BEST answer. On the answer sheet provided, shade the bubble that corresponds to the letter of your choice.

- The chemical formula of caffeine is $C_8H_{10}N_4O_2$. What is its molecular mass?
A. 165 g/mol B. 172 g/mol C. 194 g/mol D. 208 g/mol
- People usually use hydrogen peroxide (H_2O_2 ; molar mass = 34.02 g/mol) to clean their wounds. If Ricardo Delasay used 1.0 g of H_2O_2 to clean his gunshot wound, how many moles of H_2O_2 did he use?
A. 0.035 mol B. 0.030 mol C. 0.029 mol D. 0.0025 mol
- Methane (CH_4 ; molar mass = 16.04 g/mol) is one of the greenhouses gases. If 32.08 g of CH_4 was released in the atmosphere, how many molecules of CH_4 were added in the atmosphere?
A. 1.20×10^{24} B. 1.20×10^{23} C. 6.02×10^{24} D. 6.02×10^{23}

For numbers 4 and 5, consider the problem below:

- Iron (III) oxide reacts with carbon monoxide to form iron and carbon dioxide. Which is the balanced chemical equation that represents the reaction?
A. $Fe_2O_3 + 3CO \rightarrow 2Fe + 3CO_2$ C. $2Fe_2O_3 + 3CO \rightarrow 4Fe + 3CO_2$
B. $Fe_2O_3 + CO \rightarrow FeCO_2$ D. $Fe_2O_3 + 3CO \rightarrow Fe + 3CO_2$
- What mass of iron (III) oxide is needed to produce 5.6 kg of iron?
A. 4.0 kg B. 8.0 kg C. 12 kg D. 16 kg

For numbers 6-8, consider the problem below:

Ammonium carbonate decomposes to form ammonia gas, carbon dioxide and water. The reaction is represented by the equation below:



- How many moles of ammonia are formed for every mole of the reactant used?
A. 1 mole B. 2 moles C. 2.5 moles D. 3 moles
- How many moles of the reactant is required to produce 22 grams of CO_2 ?
A. 0.50 mole B. 1.0 mole C. 1.5 mole D. 2.0 moles
- How many grams of $(NH_4)_2CO_3$ is required to produce 1.5 moles of water?
A. 49 g B. 96 g C. 144 g D. 192 g

For numbers 9 and 10, consider the problem below:

45.9 g of K_2CO_3 was reacted with an excess of HCl , yielding 48.25 g of KCl .



- What is the theoretical yield?
A. 46.3 g B. 48.0 g C. 48.3 g D. 49.5 g
- What is the percent yield?
A. 92.5 % B. 93.5 % C. 97.0 % D. 97.5 %
- Which of the following will cause a soap to reduce its cleaning activity?
A. Alkyl hydrogen sulfate C. presence of surfactants
B. Hard water D. soft water
- Which of the following chemicals are linked to breast cancer in women?
A. Parabens B. PEG C. Paraffin wax D. BHA
- Which of the following disinfectants has a fungicidal property and is a potential carcinogen?

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Examanition Grade10

David Clark,Robert Talbert



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increasing resistance efforts on the part of parents teachers and the general public **The City Record** New York (N.Y.),1918 *American Universities and Colleges* ,2014-10-08 No detailed description available for American Universities and Colleges Foundations of Education Leslie S. Kaplan,William A. Owings,2022-01-21 Foundations of Education makes core topics in education accessible and personally meaningful to students pursuing a career within the education profession The Third Edition offers readers the breadth of coverage scholarly depth and conceptual analysis of contemporary issues that will help them gain a realistic and insightful perspective of the field *Standards-Based Physical Education Curriculum Development* Chair and Professor Georgia State University Atlanta Georgia Jacalyn Lund,Jacalyn Lund,Deborah Tannehill,2014-04-21 The Third Edition was created around the 2014 National Standards for Physical Education for K 12 education Written by experts with a wealth of experience designing and implementing thematic curriculum this innovative resource guides readers through the process of writing dynamic curriculum in physical education The text begins by looking at the new national standards and then examines physical education from a conceptual standpoint It goes on to examine the development of performance based assessments designed to measure the extent of student learning and explores the various curricular models common to physical education It delves into sport education adventure education outdoor education traditional multi activity fitness and movement education describing each model and how it links with physical education standards New and Key Features of the Third Edition Includes a new Chapter 2 International Perspectives on the Implementation of Standards Includes a new Chapter 4 Building the Curriculum Includes a new Chapter 6 Creating Curricular Assessments Discusses the process of designing a standards based curriculum by developing goals that are based on a sound philosophy Explores assessment and the importance of documenting students progress toward the standard Examines how teachers can provide students with opportunities to achieve their learning goals through challenging and motivating choices Scientific and Technical Aerospace Reports ,1972 Lists citations with abstracts for aerospace related reports obtained from world wide sources and announces documents that have recently been entered into the NASA Scientific and Technical Information Database **Advanced Fitness Assessment and Exercise Prescription** Ann L. Gibson,Dale R. Wagner,Vivian H. Heyward,2024-02-01 Advanced Fitness Assessment and Exercise Prescription Ninth Edition With HKPropel Online Video is the definitive resource for conducting physical fitness testing and customizing exercise programs Now in its ninth edition this comprehensive guide is fully updated with the latest research the newest exercise testing and prescription guidelines and the most up to date programming content The text reflects the most recent exercise testing and prescription guidelines from the American College of Sports Medicine ACSM along with physical activity recommendations from the U S government and American Heart Association It highlights ACSM guidelines for physical activity and exercise testing requirements to consider before beginning exercise programs Combining important research with practical application of testing and prescription protocols the ninth edition also features the following A new full color

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Table of Contents Physical Science Paper 1 Final November 2014 Examanition Grade10

1. Understanding the eBook Physical Science Paper 1 Final November 2014 Examanition Grade10
 - The Rise of Digital Reading Physical Science Paper 1 Final November 2014 Examanition Grade10
 - Advantages of eBooks Over Traditional Books
2. Identifying Physical Science Paper 1 Final November 2014 Examanition Grade10
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Physical Science Paper 1 Final November 2014 Examanition Grade10
 - User-Friendly Interface
4. Exploring eBook Recommendations from Physical Science Paper 1 Final November 2014 Examanition Grade10
 - Personalized Recommendations
 - Physical Science Paper 1 Final November 2014 Examanition Grade10 User Reviews and Ratings
 - Physical Science Paper 1 Final November 2014 Examanition Grade10 and Bestseller Lists
5. Accessing Physical Science Paper 1 Final November 2014 Examanition Grade10 Free and Paid eBooks
 - Physical Science Paper 1 Final November 2014 Examanition Grade10 Public Domain eBooks
 - Physical Science Paper 1 Final November 2014 Examanition Grade10 eBook Subscription Services
 - Physical Science Paper 1 Final November 2014 Examanition Grade10 Budget-Friendly Options

6. Navigating Physical Science Paper 1 Final November 2014 Examanition Grade10 eBook Formats
 - ePub, PDF, MOBI, and More
 - Physical Science Paper 1 Final November 2014 Examanition Grade10 Compatibility with Devices
 - Physical Science Paper 1 Final November 2014 Examanition Grade10 Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Physical Science Paper 1 Final November 2014 Examanition Grade10
 - Highlighting and Note-Taking Physical Science Paper 1 Final November 2014 Examanition Grade10
 - Interactive Elements Physical Science Paper 1 Final November 2014 Examanition Grade10
8. Staying Engaged with Physical Science Paper 1 Final November 2014 Examanition Grade10
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Physical Science Paper 1 Final November 2014 Examanition Grade10
9. Balancing eBooks and Physical Books Physical Science Paper 1 Final November 2014 Examanition Grade10
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Physical Science Paper 1 Final November 2014 Examanition Grade10
10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine Physical Science Paper 1 Final November 2014 Examanition Grade10
 - Setting Reading Goals Physical Science Paper 1 Final November 2014 Examanition Grade10
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Physical Science Paper 1 Final November 2014 Examanition Grade10
 - Fact-Checking eBook Content of Physical Science Paper 1 Final November 2014 Examanition Grade10
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
14. Embracing eBook Trends
 - Integration of Multimedia Elements

- Interactive and Gamified eBooks

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