CLASSIFYING CHEMICAL REACTIONS Name ___

Classify the following reactions as synthesis, decomposition, single replacement or double replacement.

8.
$$CaCl_2 + F_2 \rightarrow CaF_2 + Cl_2$$

10.
$$N_2 + 3H_2 \rightarrow 2NH_3$$

12.
$$(NH_4)_2SO_4 + Ba(NO_3)_2 \rightarrow BaSO_4 + 2NH_4NO_3$$

13.
$$MgI_2 + Br_2 \rightarrow MgBr_2 + I_2$$

14.
$$SO_3 + H_2O \rightarrow H_2SO_4$$

15.
$$6KCI + Zn_3(PO_4)_2 \rightarrow 3ZnCI_2 + 2K_3PO_4$$

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SJ Ball

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