

Element	symbol	Atomic number	Electronic arrangement in different orbit					
			K	L	M	N	O	P
Hydrogen	H	1	1					
Helium	He	2	2					
Lithium	Li	3	2	1				
Beryllium	Be	4	2	2				
Boron	B	5	2	3				
Carbon	C	6	2	4				
Nitrogen	N	7	2	5				
Oxygen	O	8	2	6				
Fluorine	F	9	2	7				
Neon	Ne	10	2	8				
Sodium	Na	11	2	8	1			
Magnesium	Mg	12	2	8	2			
Aluminium	Al	13	2	8	3			
Silicon	Si	14	2	8	4			
Phosphorus	P	15	2	8	5			
Sulphur	S	16	2	8	6			
Chlorine	Cl	17	2	8	7			
Argon	Ar	18	2	8	8			
Potassium	K	19	2	8	8	1		
Calcium	Ca	20	2	8	8	2		

● $c = v\lambda$

● $\bar{v} = \frac{1}{\lambda} = \frac{v}{c}$

[c = velocity of light in free space,

v = frequency, λ = wavelength]

[\bar{v} = wave number]

Physical Science Notes Chapter Atomic Structure

Marcel A. Müller



Physical Science Notes Chapter Atomic Structure:

Decoding **Physical Science Notes Chapter Atomic Structure**: Revealing the Captivating Potential of Verbal Expression

In an era characterized by interconnectedness and an insatiable thirst for knowledge, the captivating potential of verbal expression has emerged as a formidable force. Its power to evoke sentiments, stimulate introspection, and incite profound transformations is genuinely awe-inspiring. Within the pages of "**Physical Science Notes Chapter Atomic Structure**," a mesmerizing literary creation penned by a celebrated wordsmith, readers embark on an enlightening odyssey, unraveling the intricate significance of language and its enduring impact on our lives. In this appraisal, we shall explore the book's central themes, evaluate its distinctive writing style, and gauge its pervasive influence on the hearts and minds of its readership.

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