

Density Worksheet

Physical Science

$D = m/V$

| Densities of Common Substances @ 20°C | | | | |
|---------------------------------------|------------------------------|--|-----------|------------------------------|
| Substance | Density (g/cm ³) | | Substance | Density (g/cm ³) |
| Oxygen | 0.00133 | | Aluminum | 2.70 |
| Hydrogen | 0.000084 | | Iron | 7.87 |
| Ethanol | 0.785 | | Copper | 8.96 |
| Benzene | 0.880 | | Silver | 10.5 |
| Water | 1.000 | | Lead | 11.34 |
| Magnesium | 1.74 | | Mercury | 13.6 |
| Salt (sodium chloride) | 2.16 | | Gold | 19.32 |

1. The ratio of an object's mass to its _____ is called the *density* of the object.
2. A kilogram of lead occupies a much smaller volume than a kilogram of water, because _____ has a much higher *density*.
3. For the masses and volumes indicated, calculate the **density** in grams per cubic centimeters.
 - a. mass = 453 g; volume = 225 cm³
 - b. mass = 5.0 g; volume = 10.0 cm³
 - c. mass = 26.1 g; volume = 2.0 mL
4. If 89.2 mL of a liquid has a mass of 75.2 g, calculate the liquid's density.
5. A cube of metal weighs 1450 g and displaces 542 mL of water when immersed. Calculate the density of the metal.

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Denise Eby, Robert B. Horton



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Table of Contents Physical Science If8767 Page 99 Answers

1. Understanding the eBook Physical Science If8767 Page 99 Answers
 - The Rise of Digital Reading Physical Science If8767 Page 99 Answers
 - Advantages of eBooks Over Traditional Books
2. Identifying Physical Science If8767 Page 99 Answers
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Physical Science If8767 Page 99 Answers
 - User-Friendly Interface
4. Exploring eBook Recommendations from Physical Science If8767 Page 99 Answers
 - Personalized Recommendations
 - Physical Science If8767 Page 99 Answers User Reviews and Ratings
 - Physical Science If8767 Page 99 Answers and Bestseller Lists
5. Accessing Physical Science If8767 Page 99 Answers Free and Paid eBooks
 - Physical Science If8767 Page 99 Answers Public Domain eBooks
 - Physical Science If8767 Page 99 Answers eBook Subscription Services
 - Physical Science If8767 Page 99 Answers Budget-Friendly Options
6. Navigating Physical Science If8767 Page 99 Answers eBook Formats

- ePub, PDF, MOBI, and More
- Physical Science If8767 Page 99 Answers Compatibility with Devices
- Physical Science If8767 Page 99 Answers Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Physical Science If8767 Page 99 Answers
 - Highlighting and Note-Taking Physical Science If8767 Page 99 Answers
 - Interactive Elements Physical Science If8767 Page 99 Answers
- 8. Staying Engaged with Physical Science If8767 Page 99 Answers
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Physical Science If8767 Page 99 Answers
- 9. Balancing eBooks and Physical Books Physical Science If8767 Page 99 Answers
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Physical Science If8767 Page 99 Answers
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Physical Science If8767 Page 99 Answers
 - Setting Reading Goals Physical Science If8767 Page 99 Answers
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Physical Science If8767 Page 99 Answers
 - Fact-Checking eBook Content of Physical Science If8767 Page 99 Answers
 - Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
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 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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